

The XXII International Baldin Seminar on High Energy Physics Problems

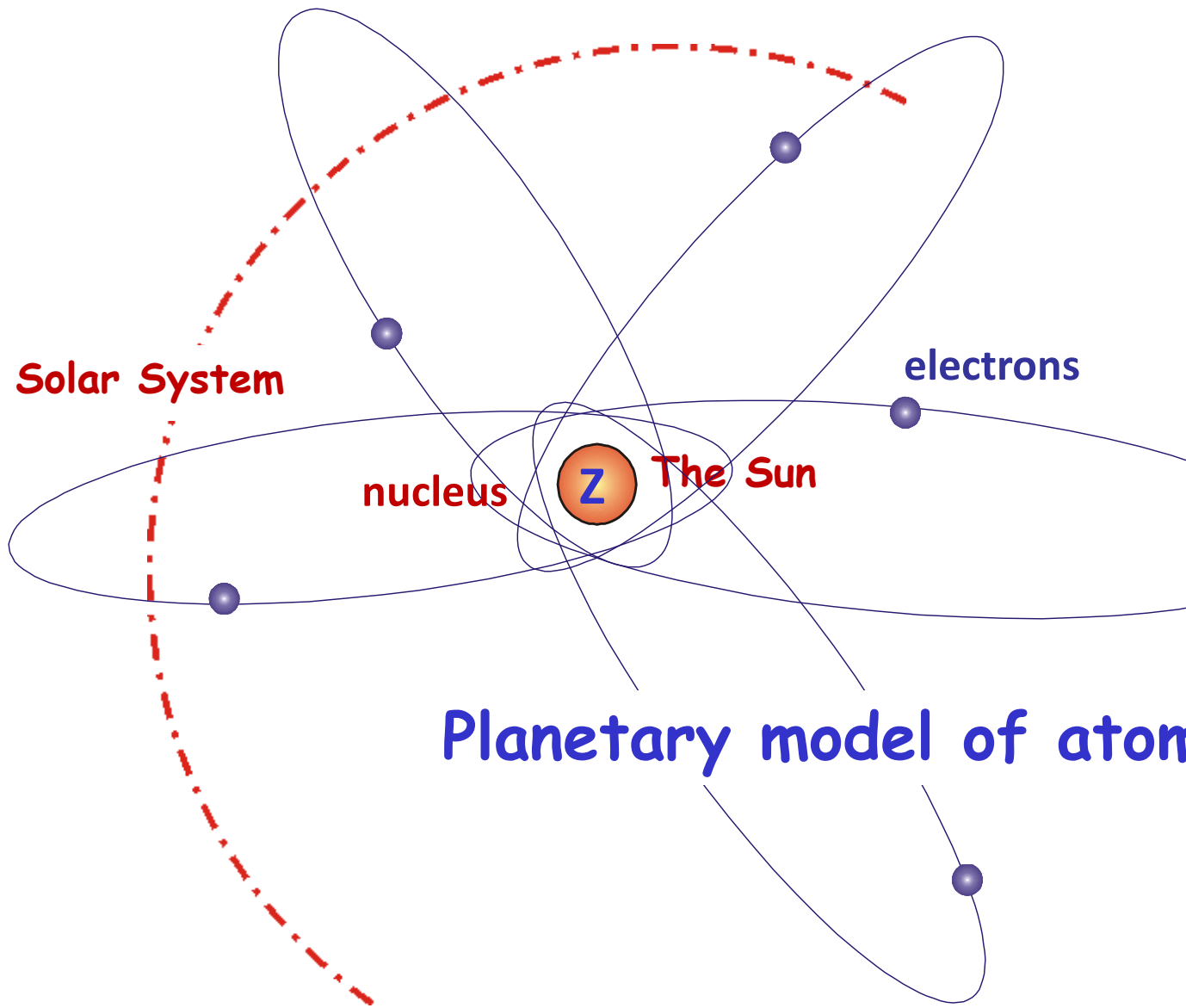
At the End of the Nuclear Map

Yuri Oganessian

Flerov Laboratory of Nuclear Reactions
Joint Institute for Nuclear Research

141980 DUBNA, Moscow region, RF

"Relativistic Nuclear Physics and Quantum Chromodynamics"
JINR September 15-20, 2014 in Dubna, Russia.



E. Rutherford 1911

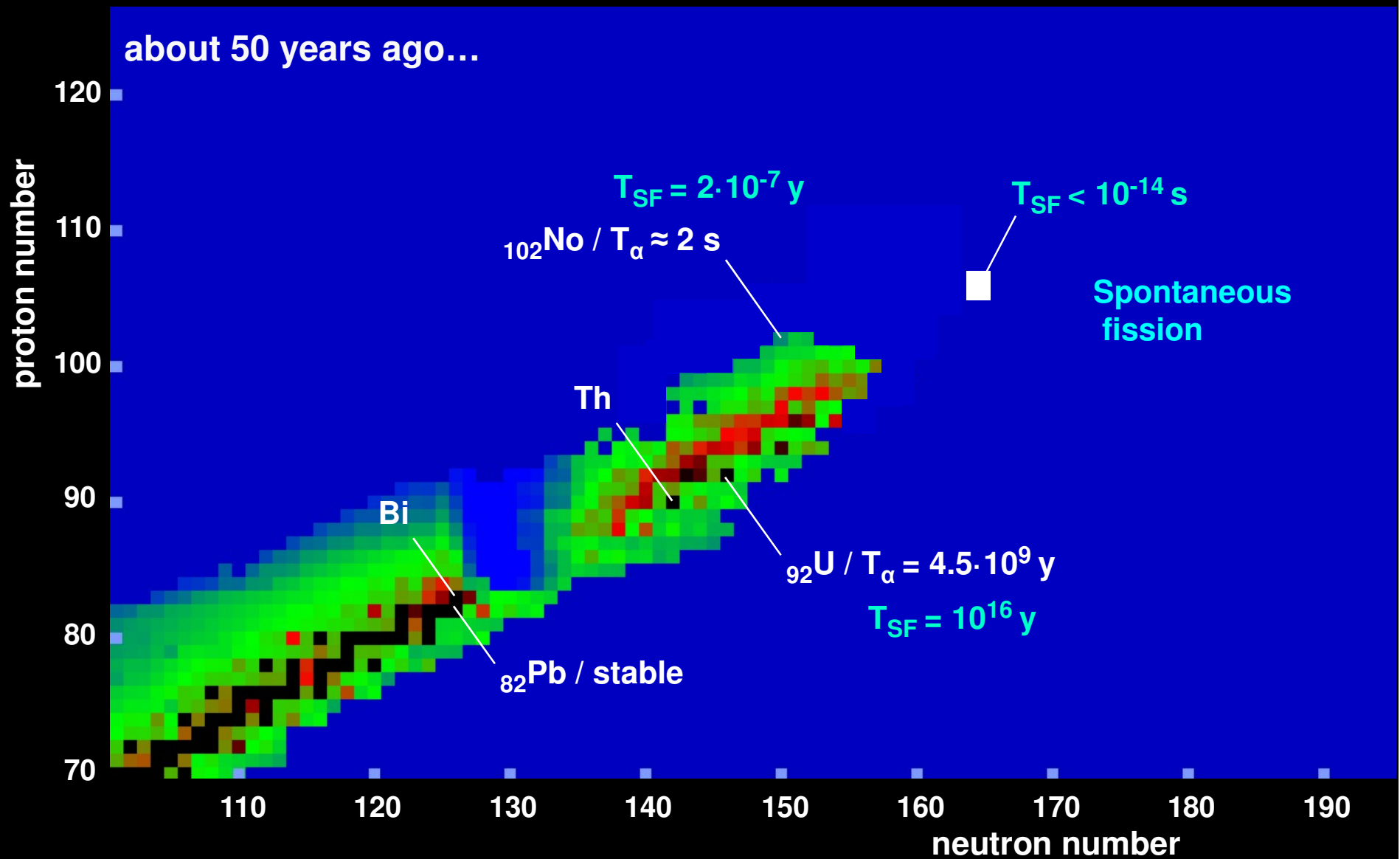
Planetary model of atom

Today we will discuss:

- **How big a nucleus may be,**
- **What is a maximum number of protons and neutrons it may contain,**
- **What is the limit of atomic nuclei mass and how it is determined.**

Chart of nuclides

Extrapolation from the decay properties of transuranium nuclei



In the first attempts of describing the properties of nuclear matter a daring supposition was made that atomic nucleus is an object similar to a drop of positively charged liquid, the so called



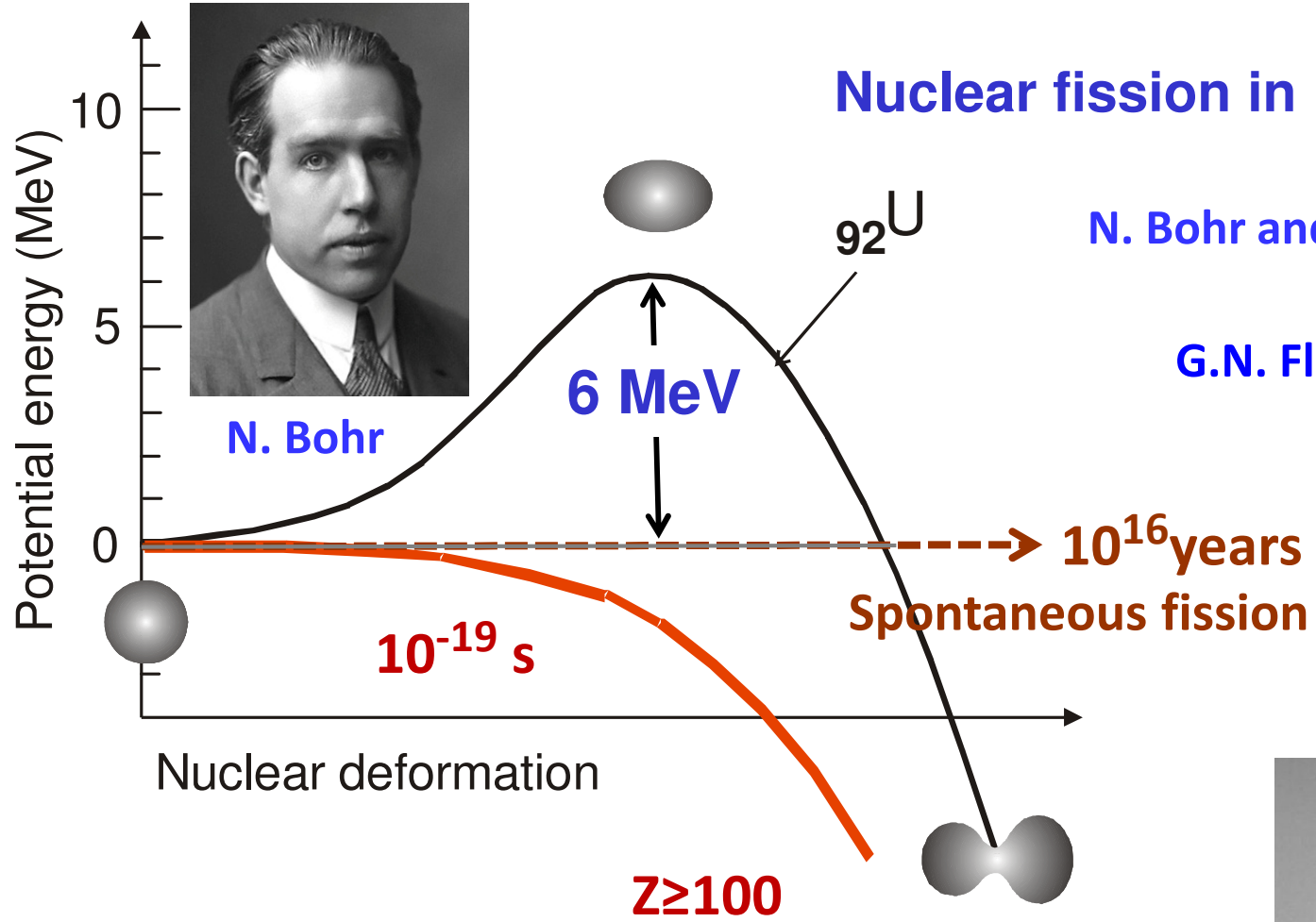
G. Gamow 1928

Nuclear Charge Liquid-Drop Model

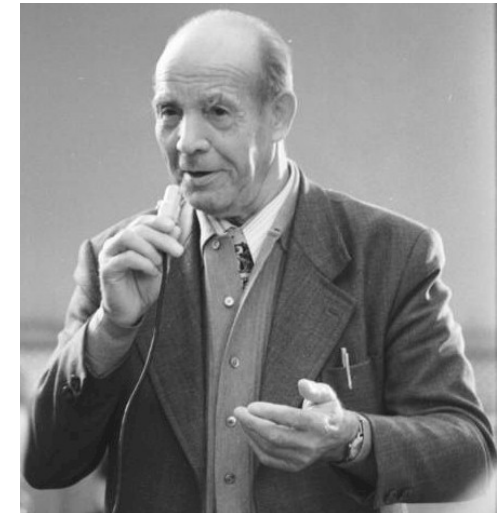
Nuclear fission in Liquid Drop Model

N. Bohr and J. A. Wheeler

G.N. Flerov

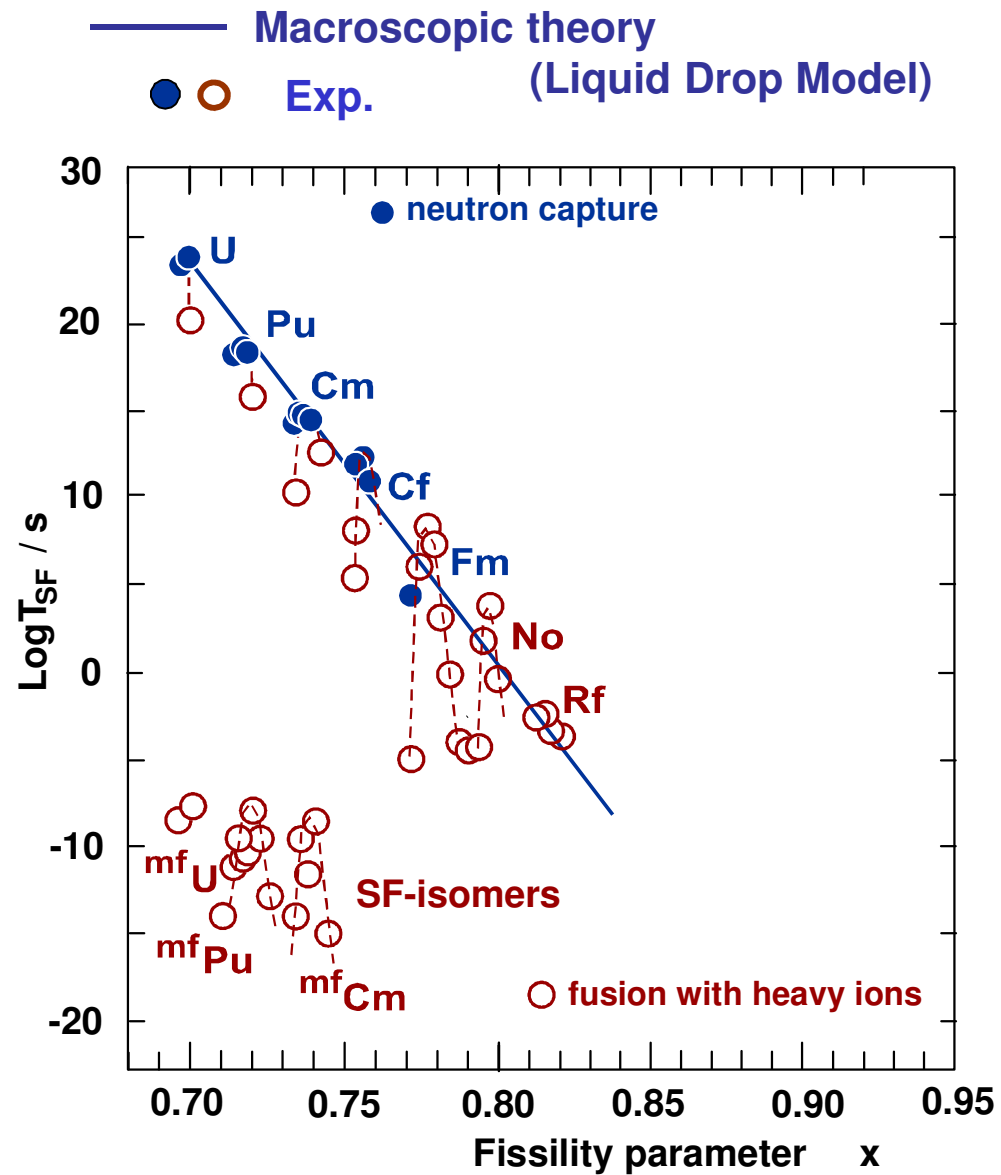
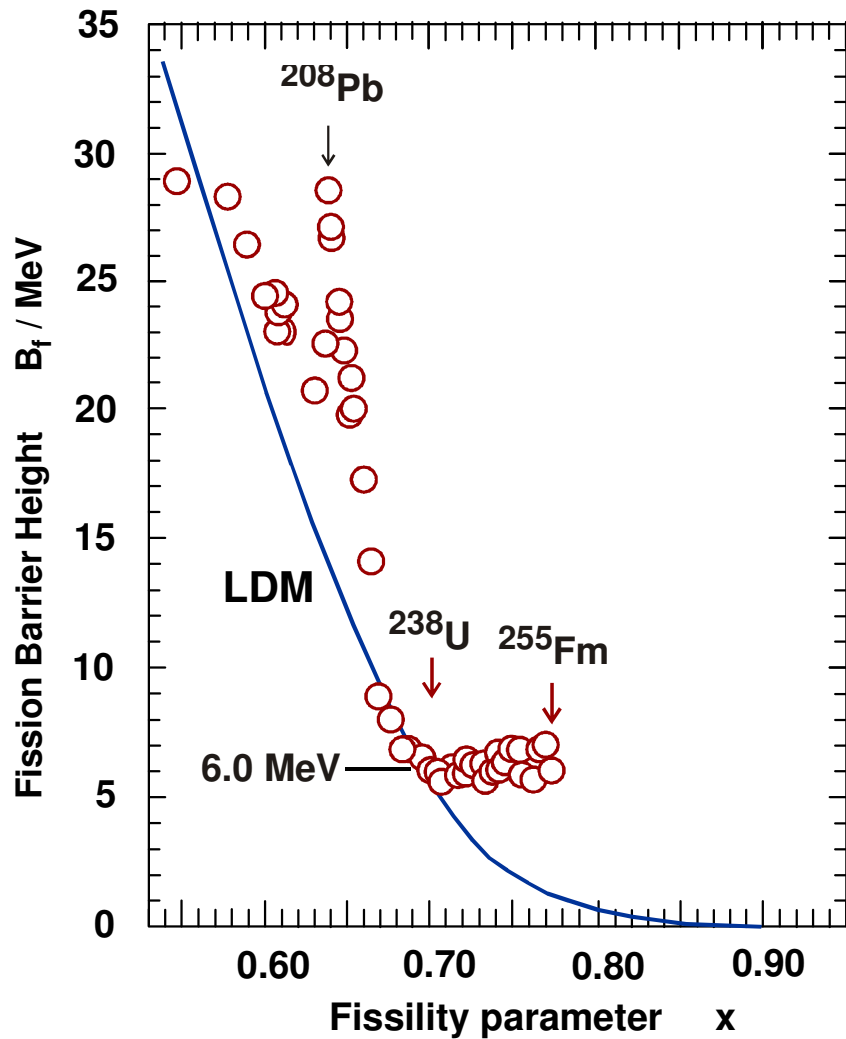


N. Bohr

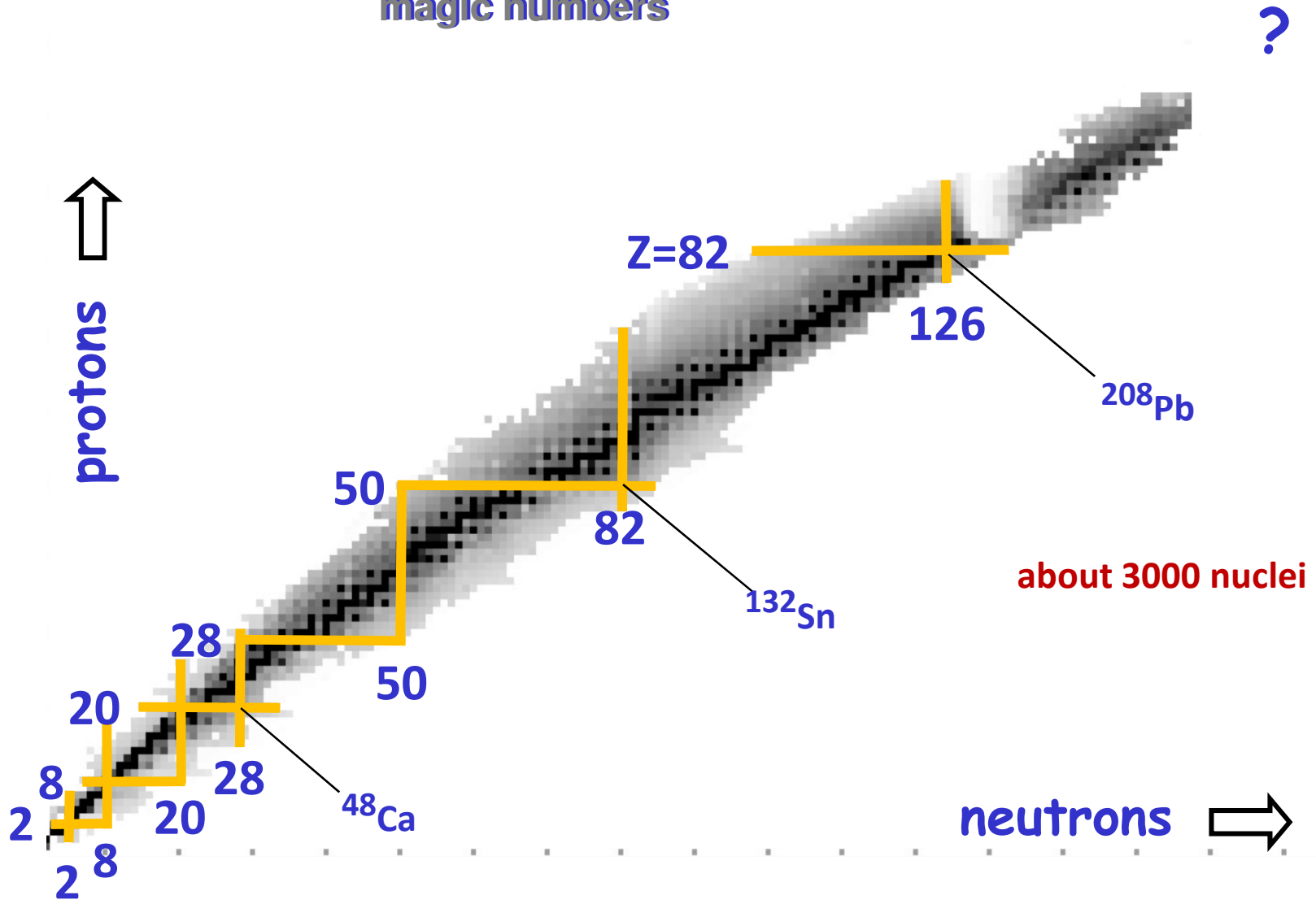


K.A. Petrzhak

Spontaneous Fission



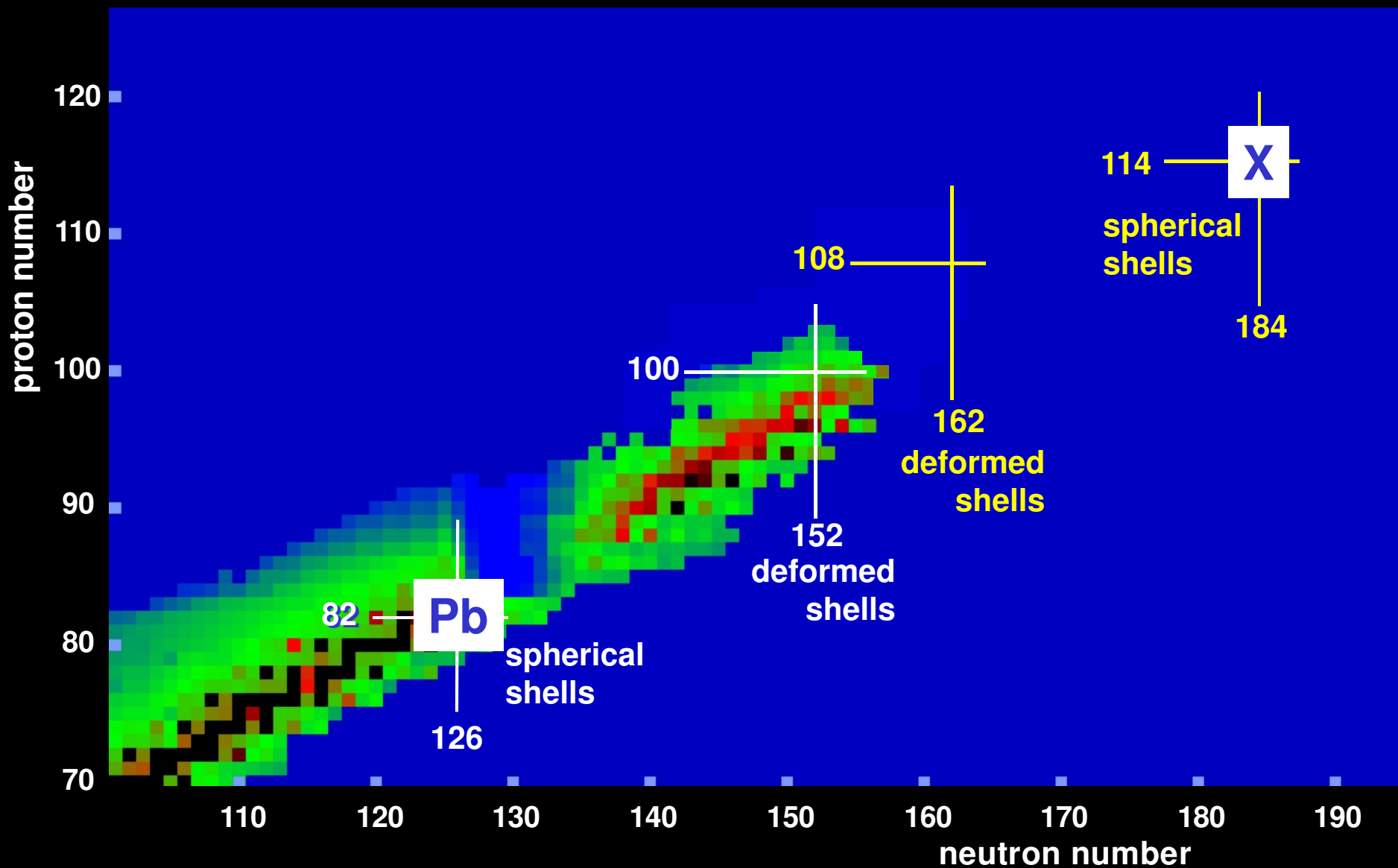
Shells and magic numbers

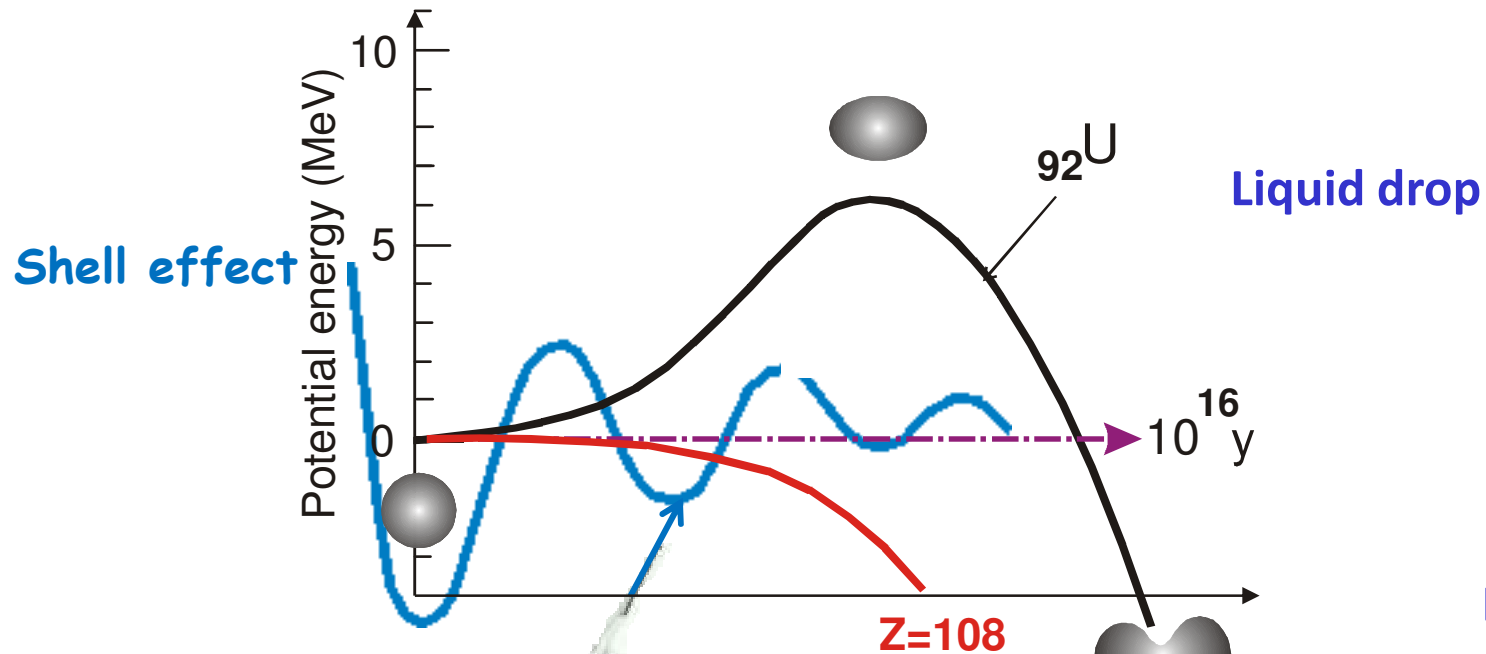


Yu/ Oganessian. XXII International Baldin Seminar, September 15-20, 2014, JINR, Dubna

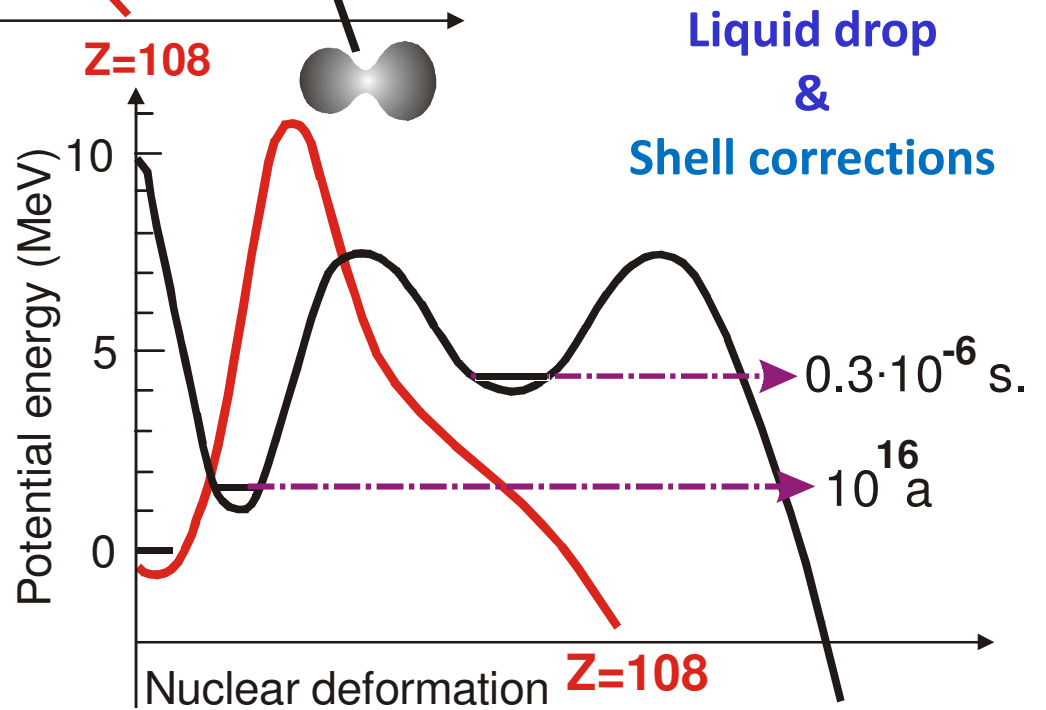
Chart of nuclides

Nuclear shells (macro-microscopic approach)



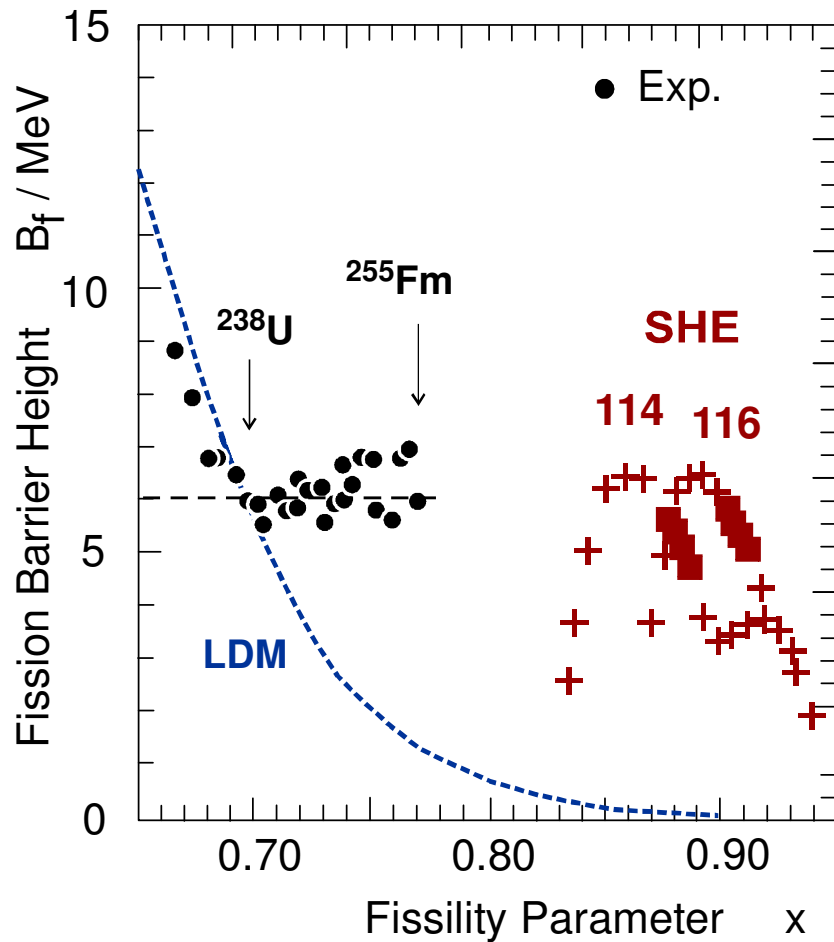


V.M. Strutinsky 1967

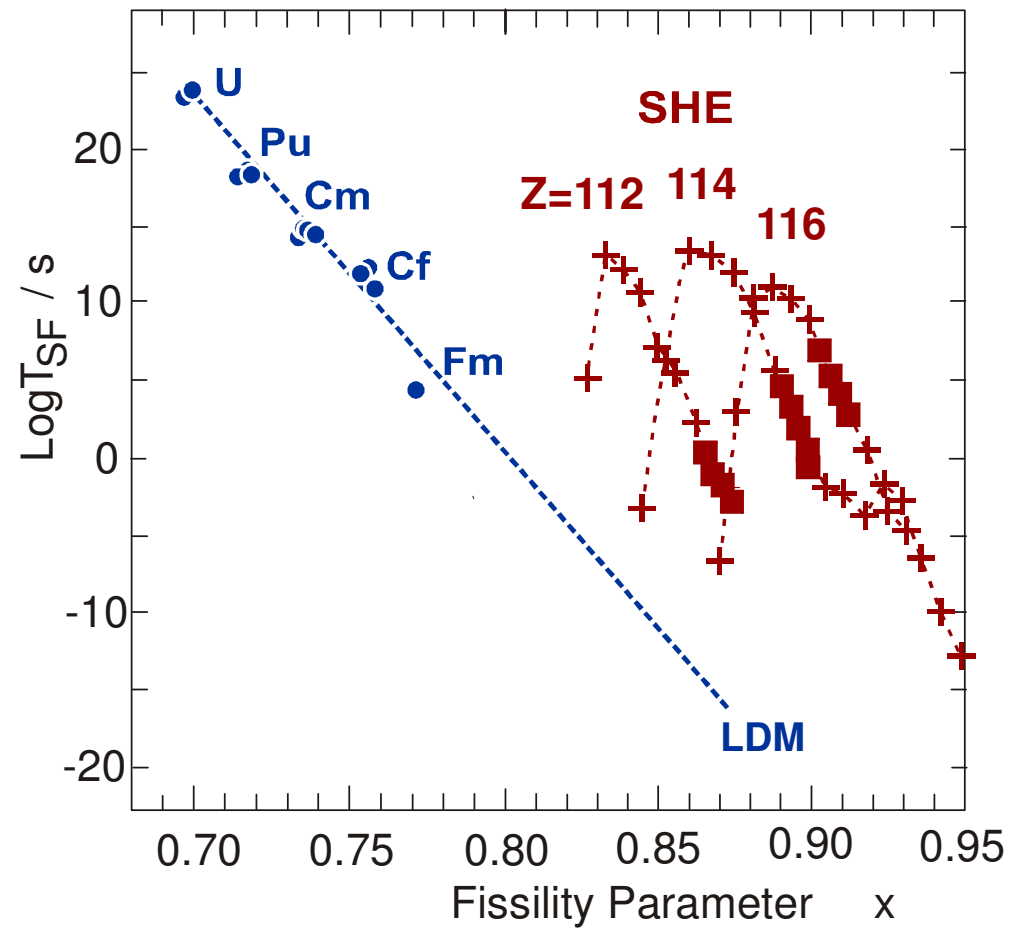


Predictions of the microscopic theory

Fission Barriers

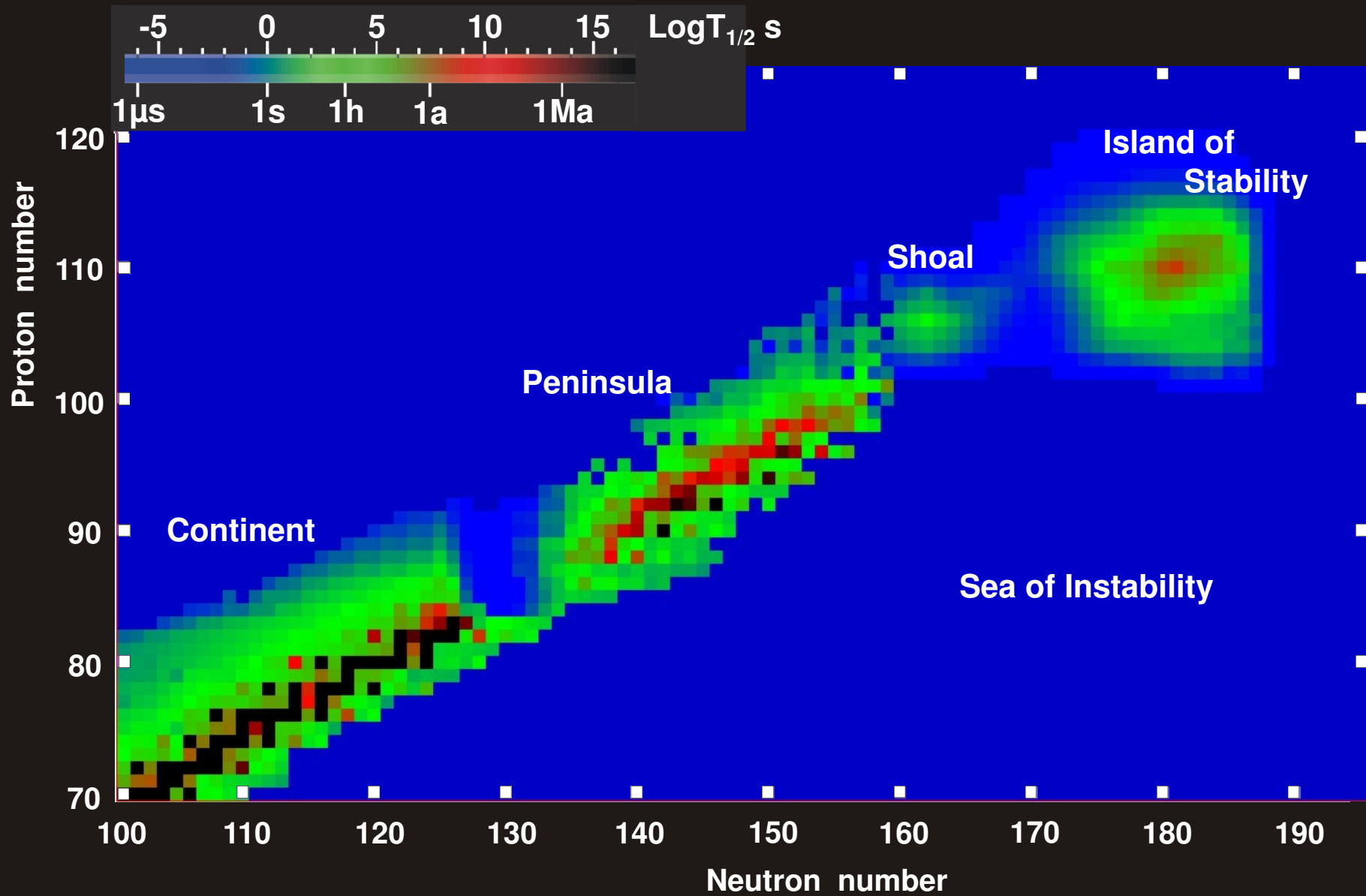


...and Half - Lives



New lands

Microscopic theory

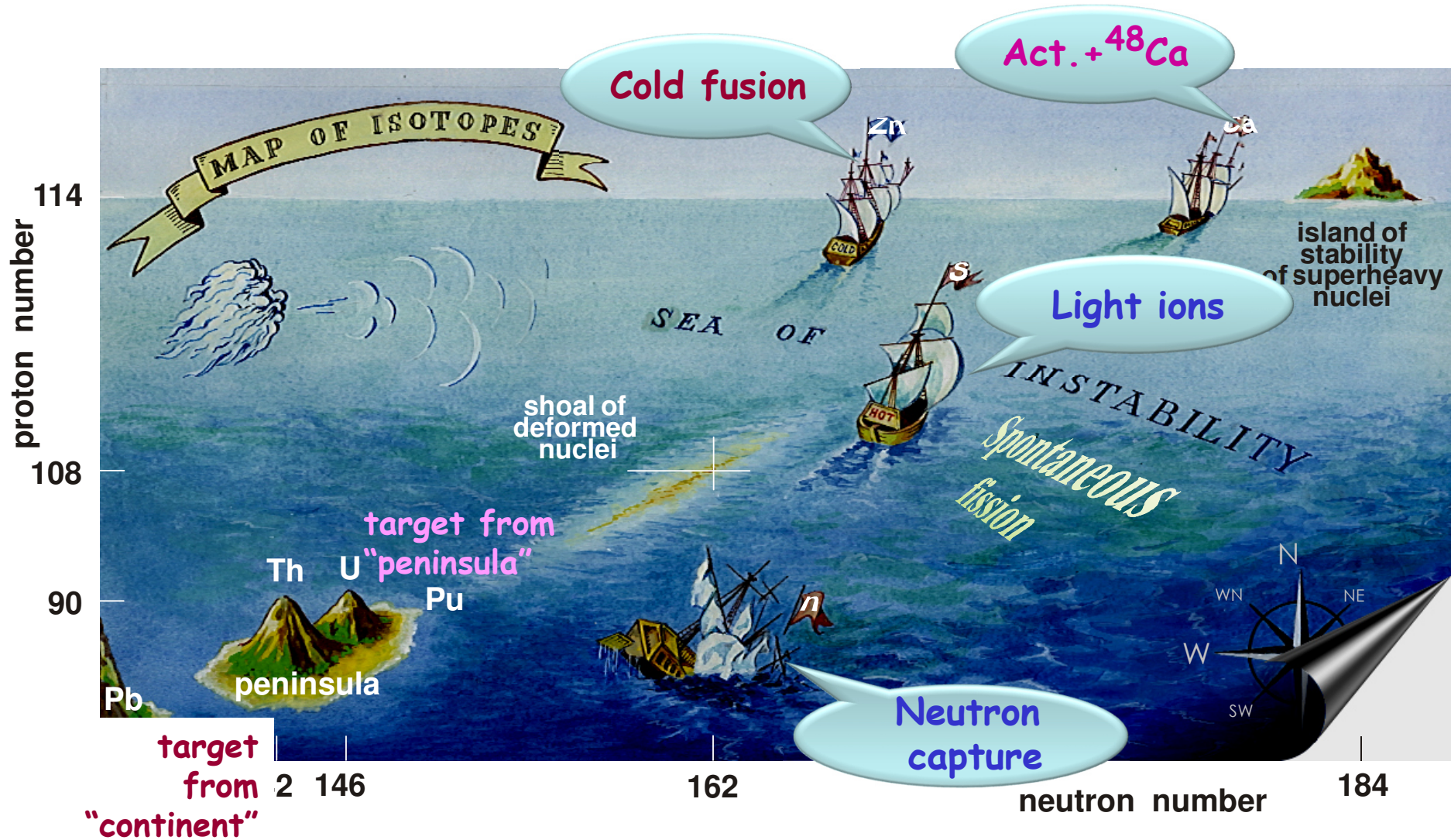


Yu. Oganessian 2012

Reaction of Synthesis

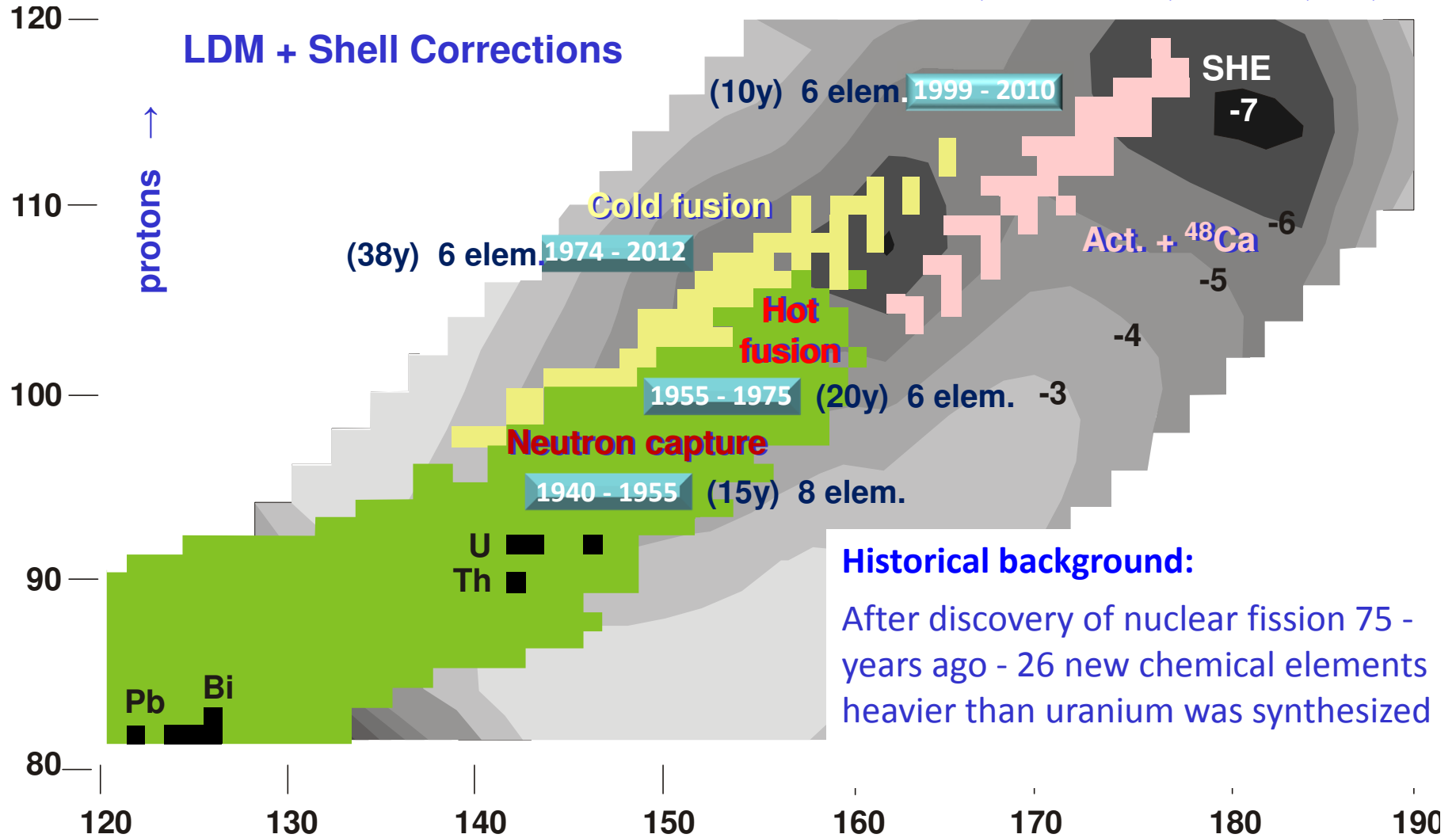
In laboratory conditions the heaviest elements are produced in collisions of massive nuclei at great velocities... (up to 1/10 of the speed of light)

Reactions of synthesis



Reactions of Synthesis

A. Sobiczewski, K. Pomorski, PPNP 58, 292, 2007



hot fusion

^{48}Ca -induced reactions

$E_x = 35 - 45 \text{ MeV}$

$x = 3 - 4$

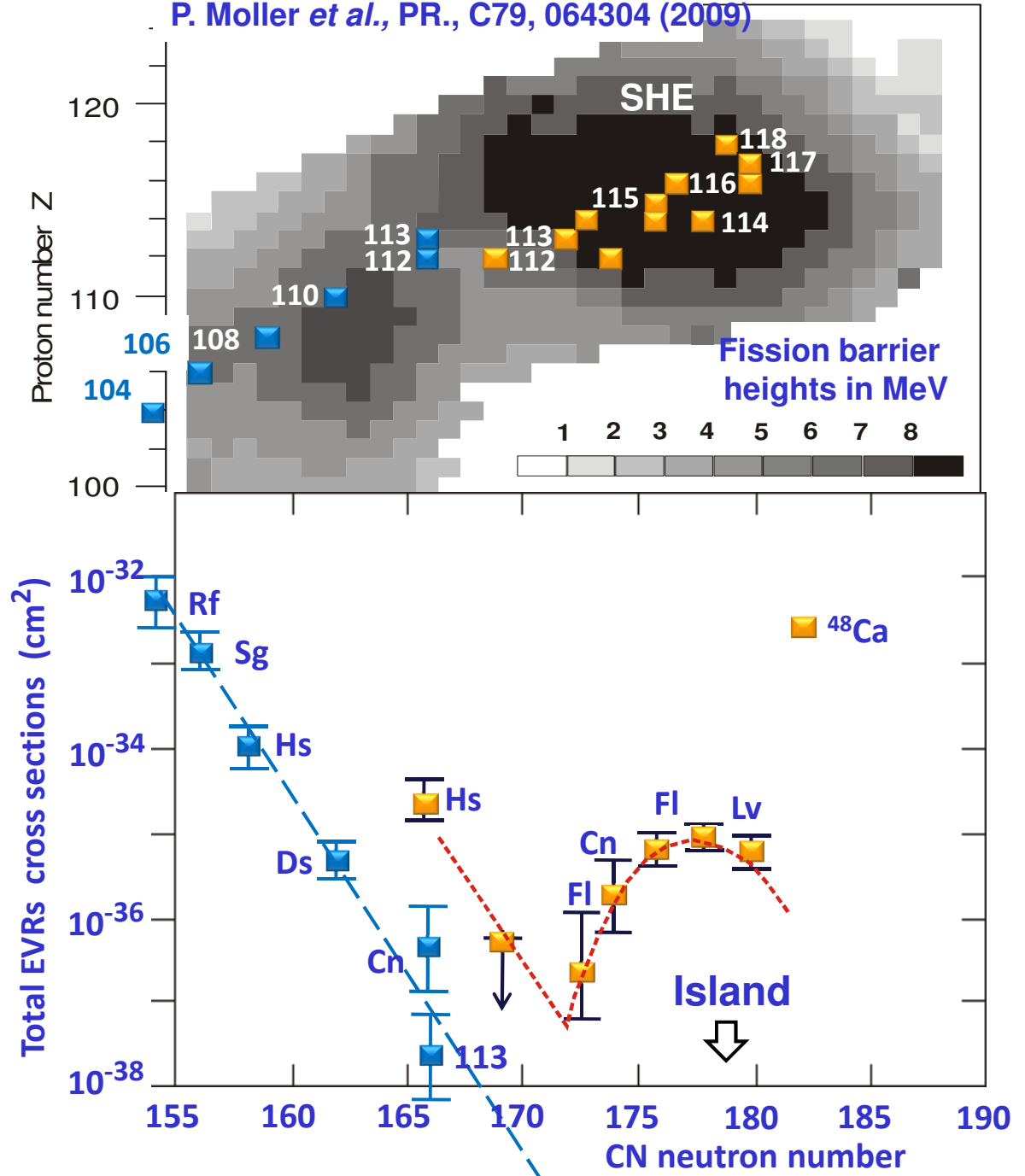
cold fusion

$^{208}\text{Pb}, ^{209}\text{Bi} + ^{50}\text{Ti}, \dots, ^{70}\text{Zn}$

$E_x = 12 - 15 \text{ MeV}$

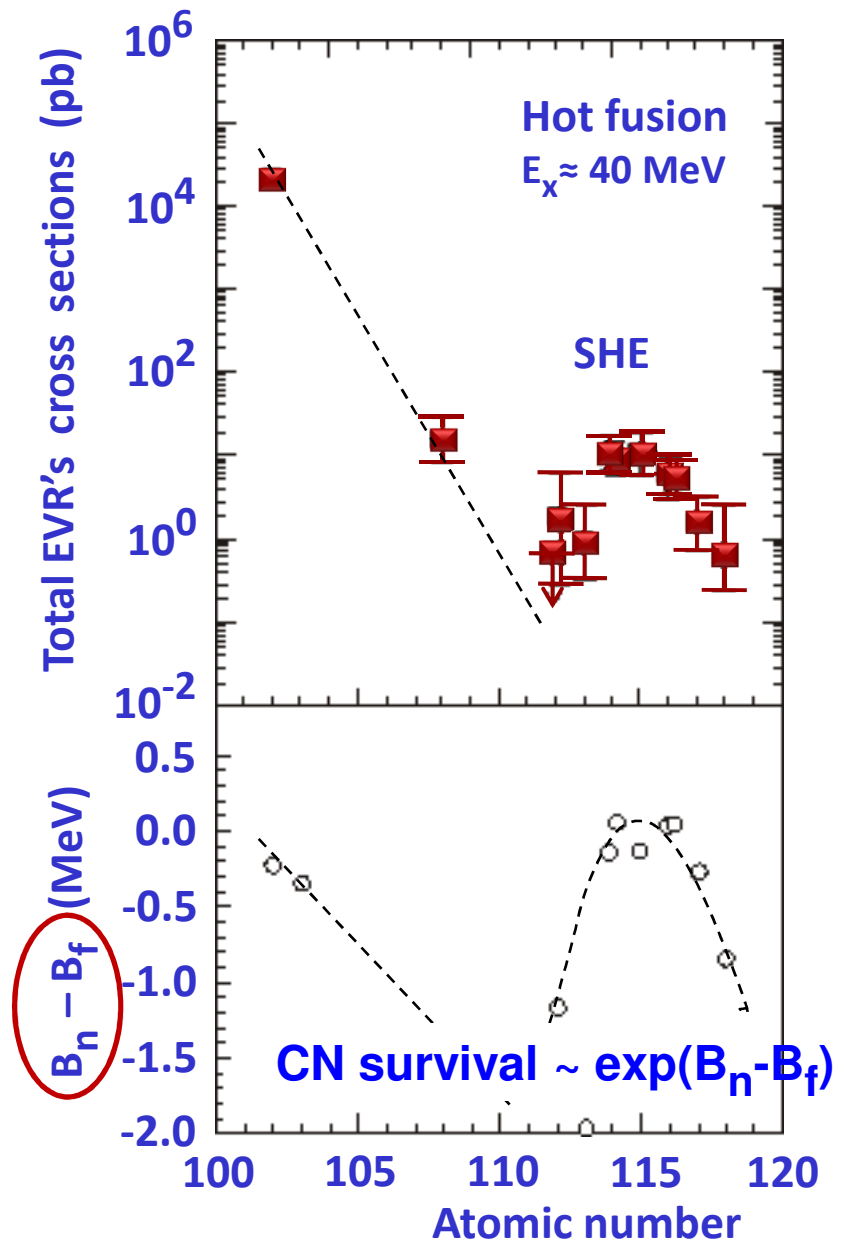
$x = 1$

P. Moller et al., PR., C79, 064304 (2009)



$$P_{sur} \sim P_{xn} \cdot \prod_{i=1}^x (\Gamma_{ni} / \Gamma_{fi})^x$$

$$\Gamma_n / \Gamma_f \sim e^{-\frac{B_f - B_n}{T}}$$



TARGETS

48Ca - PROJECTILES

Energy:

235-250 MeV

Intensity:

1.0-1.2 pμA

Consumption:

0.5 mg/h

Beam dose:

up to $4.5 \cdot 10^{19}$

Isotope	Target thickness mg/cm ²	Isotope enrichment %	Setup
233U	0.44	99.92	DGFRS
237Np	0.35	99.3	DGFRS
238U	0.35	99.3	DGFRS
242Pu	0.40	99.98	DGFRS
	1.40	99.98	Chem.
243Am	0.36	99.9	DGFRS
	1.20	99.9	Chem.
244Pu	0.38	98.6	DGFRS
245Cm	0.35	98.7	DGFRS
248Cm	0.35	97.4	DGFRS
249Bk	0.35	≥90	DGFRS
249Cf	0.30	≥90	DGFRS

1998 - 2007

Experimental technique

Dubna Gas-Filled Recoil Separator

Transmission for:

EVR 35-40%

target-like 10^{-4} - 10^{-7}

projectile-like 10^{-15} - 10^{-17}

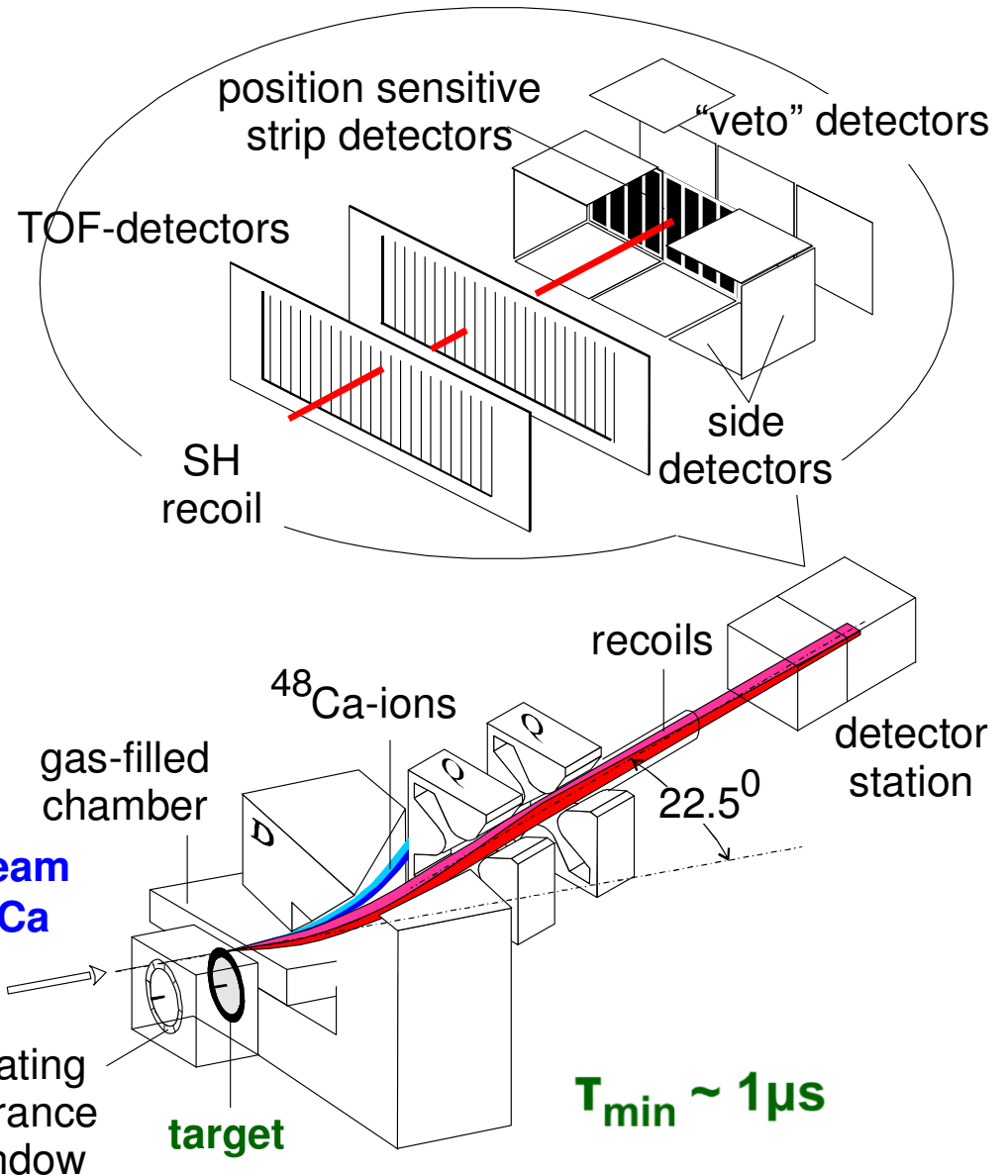
Registration efficiency:

for α -particles 87%

for SF

single fragment 100%

two fragments $\approx 40\%$



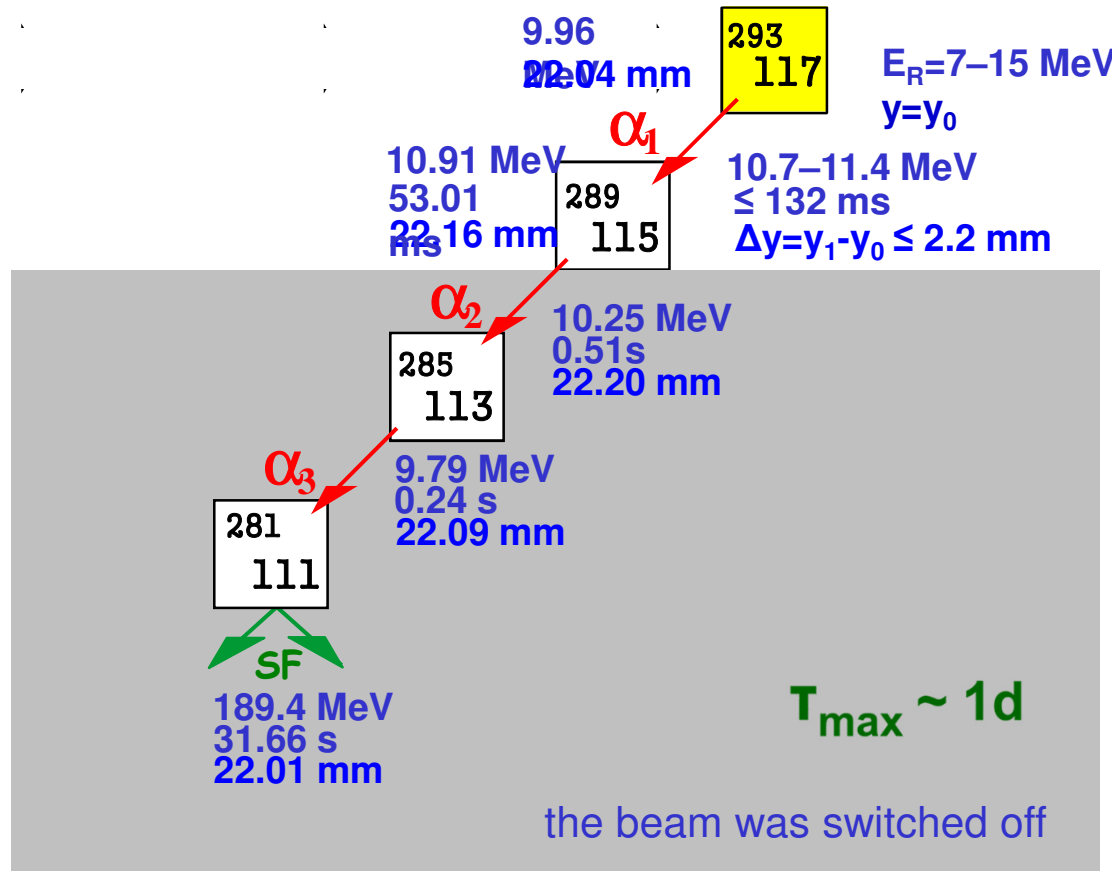
low-background detection scheme

Yu. Oganessian 2010

^{249}Bk
target

separator

focal plane
detector array



Even-Z Nuclei

Z=118

116

114

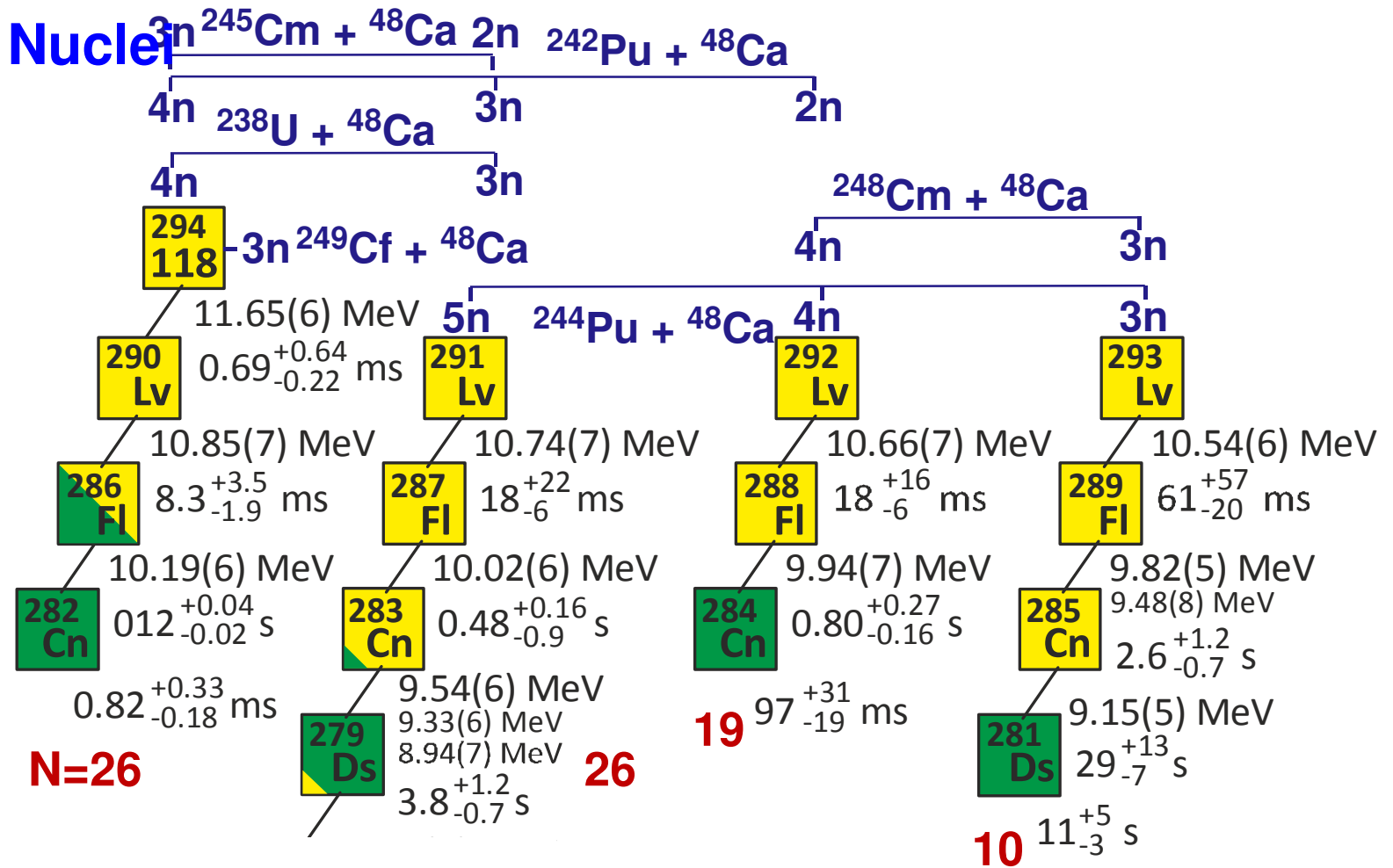
112

110

108

106

104



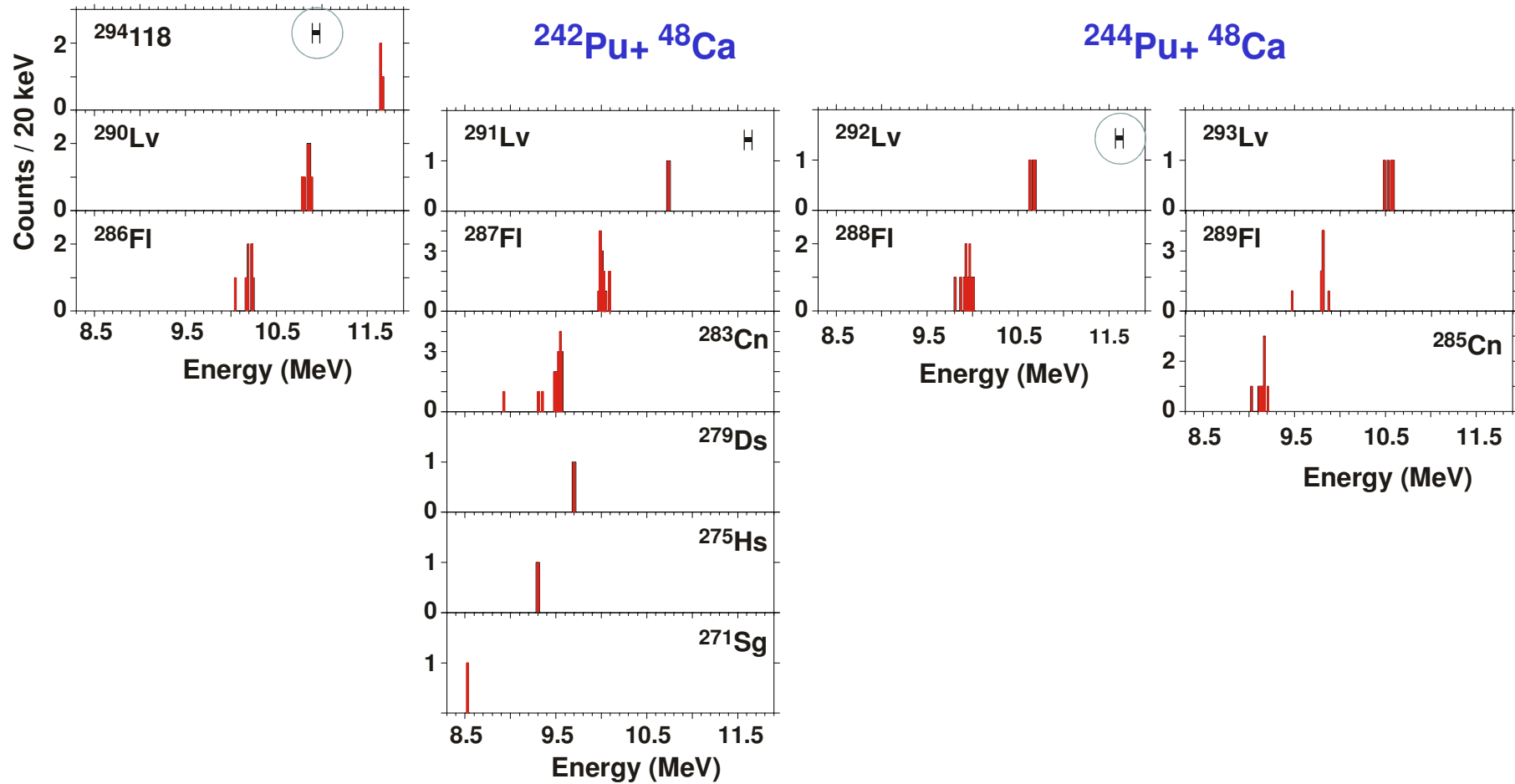
Decay chains

Even Z Nuclei

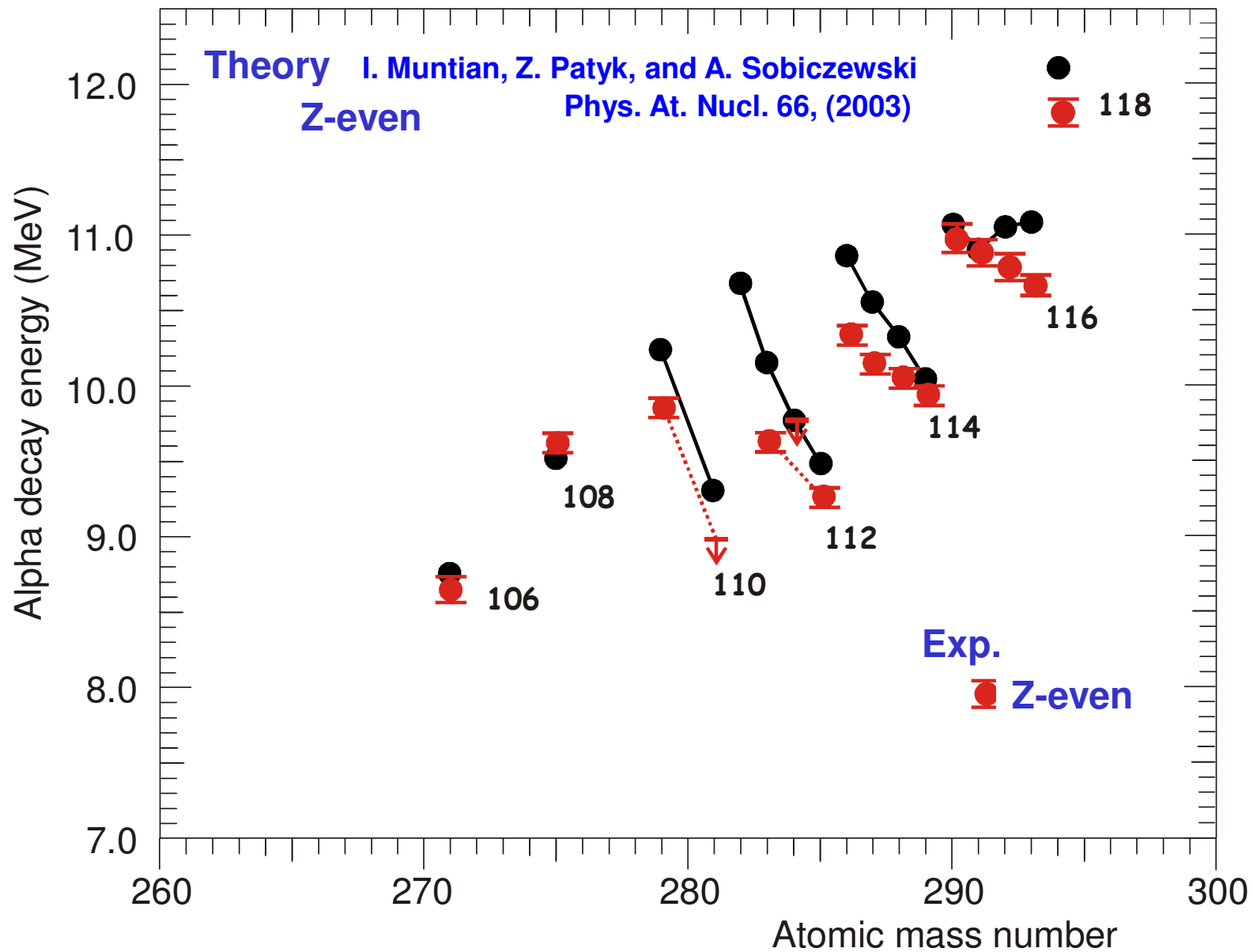
1999 - 2005

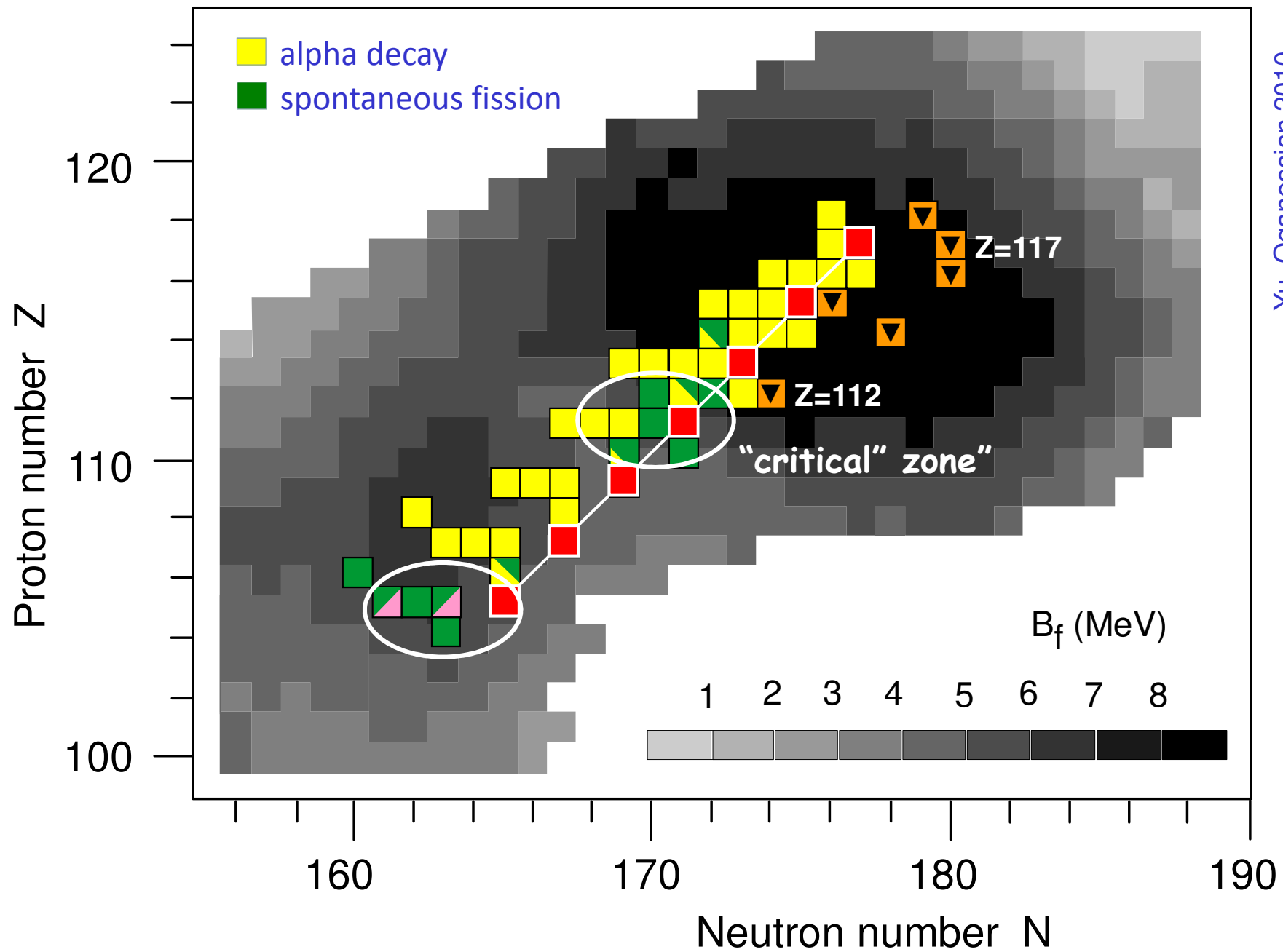
$^{249}\text{Cf} + ^{48}\text{Ca}$

Energy spectra of alpha particles



Alpha - decay

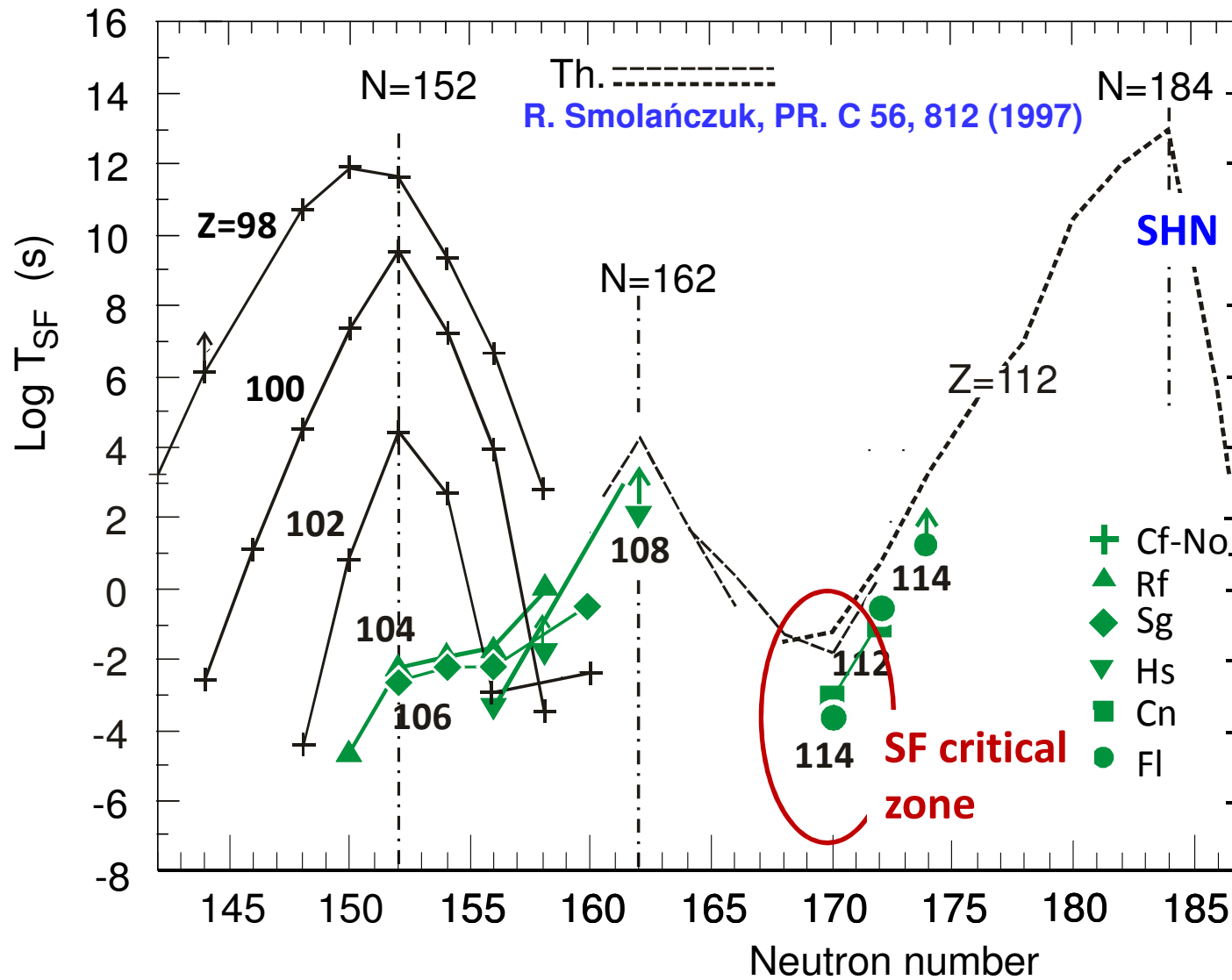


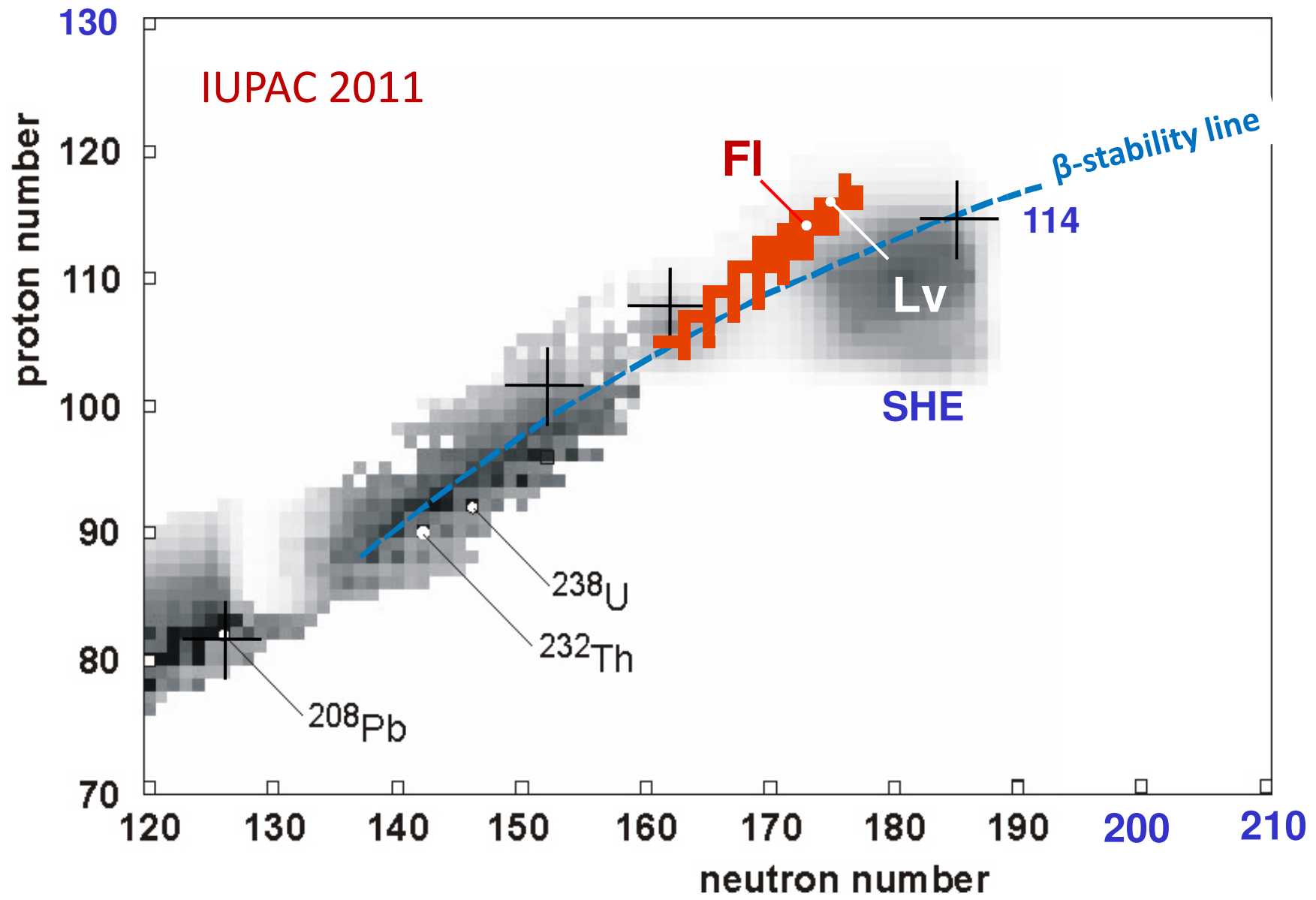


Yu. Oganessian 2010

Spontaneous fission

even-even isotopes



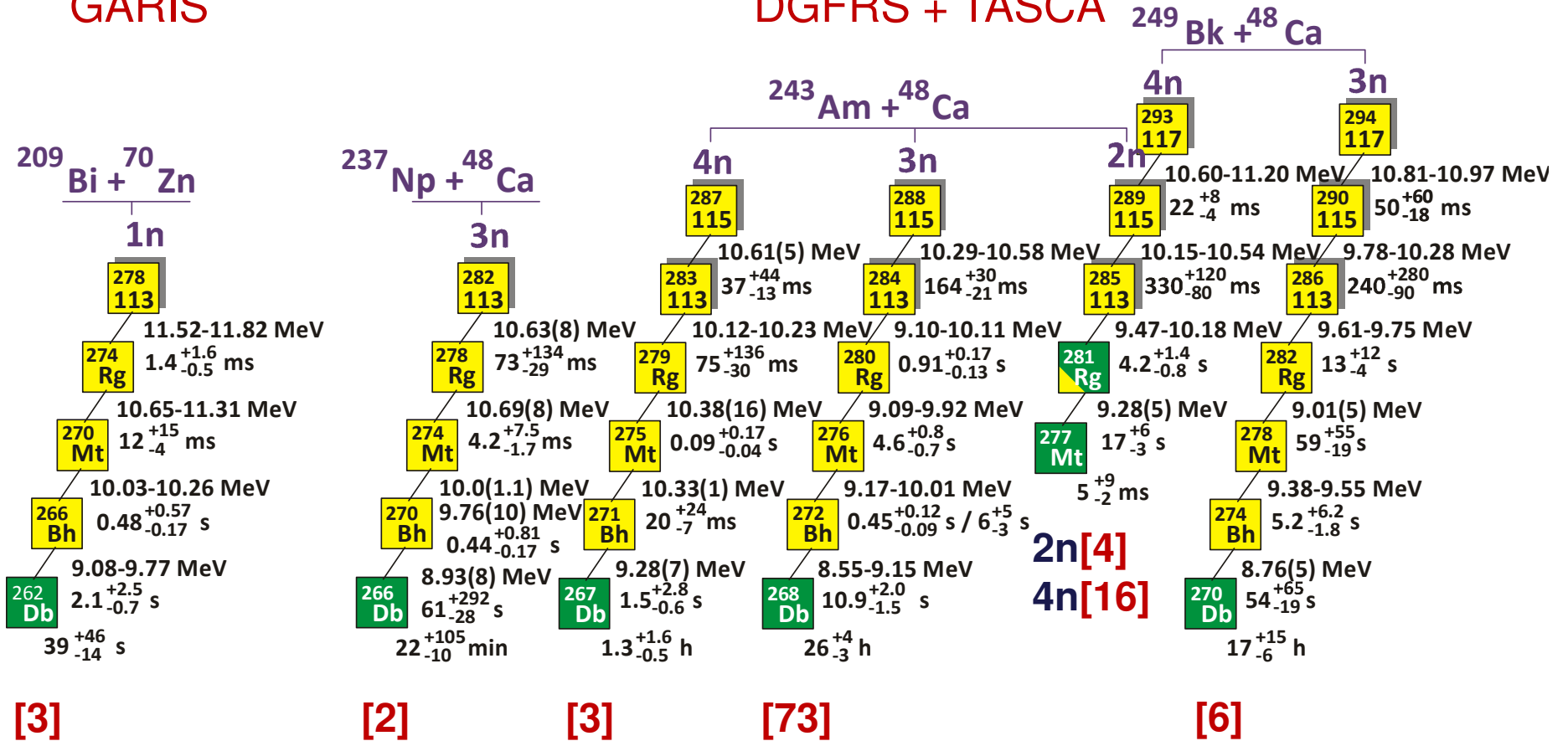


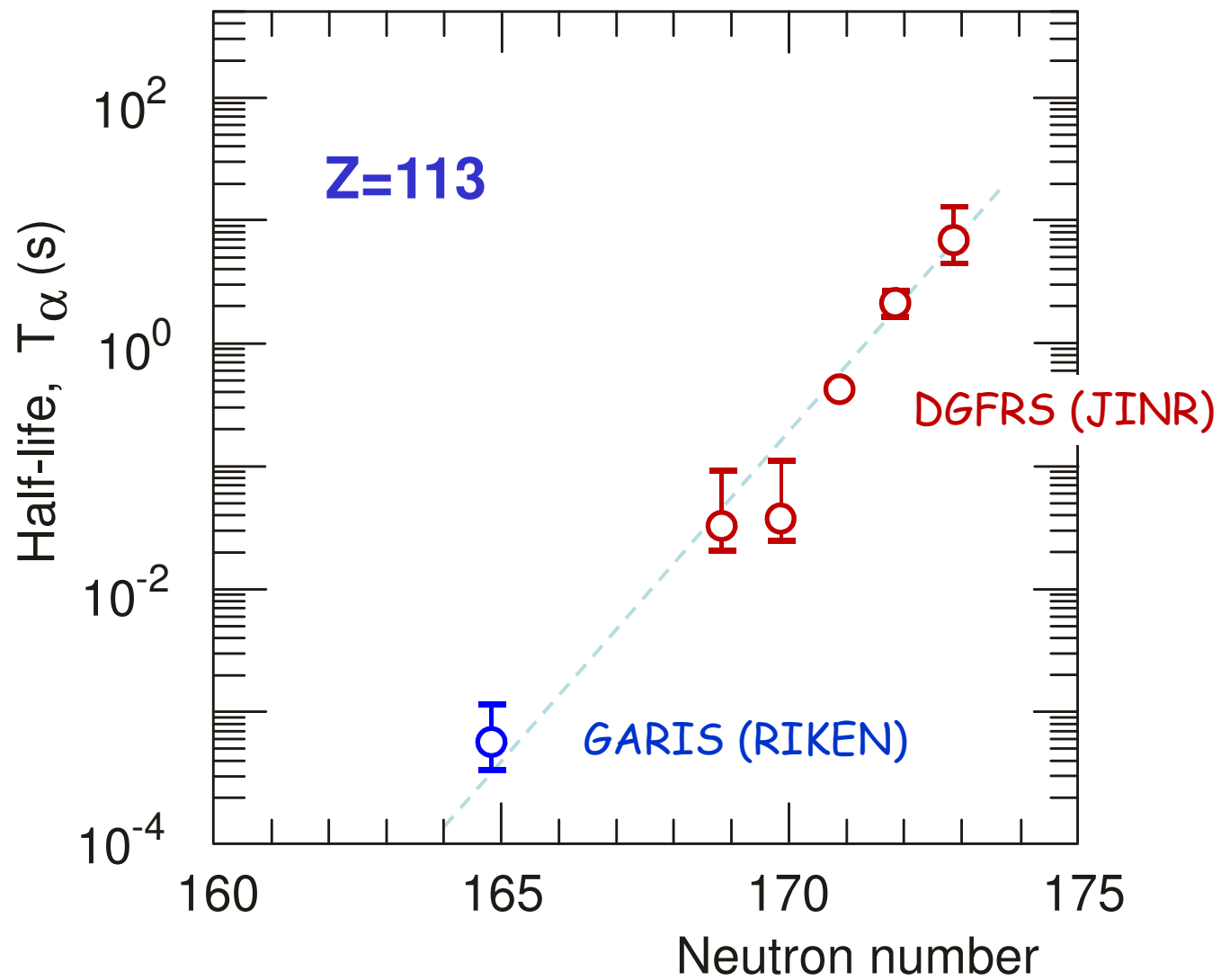
Yu/ Oganessian. XXII International Baldin Seminar, September 15-20, 2014, JINR, Dubna

June, 2013

GARIS

DGFRS + TASCA

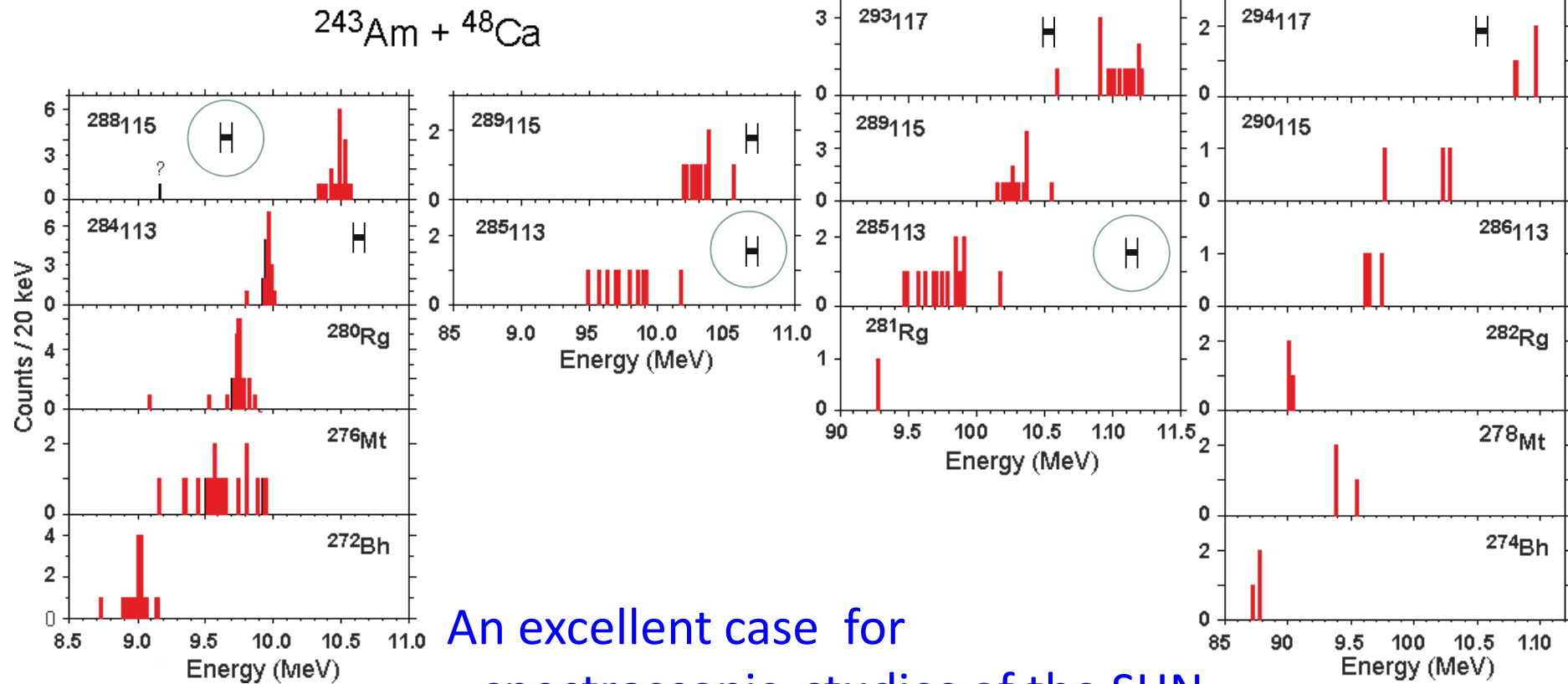




Odd Z Nuclei

2003 - 2012

$^{249}\text{Bk} + ^{48}\text{Ca}$

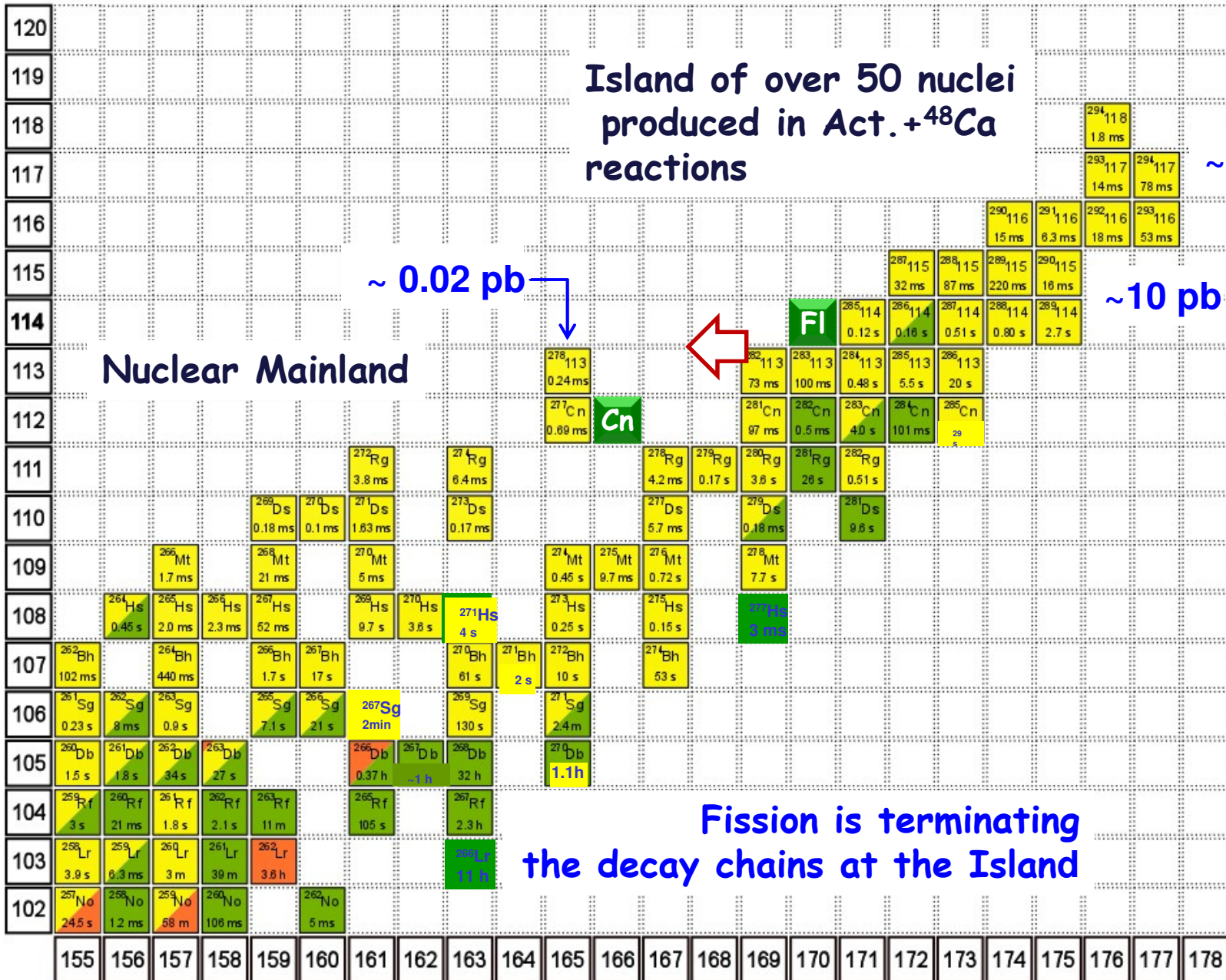


An excellent case for spectroscopic studies of the SHN in $\langle\alpha\text{-}\gamma\rangle$ coincidence experiments

Confirmations of DGFRS data

2007 - 2014

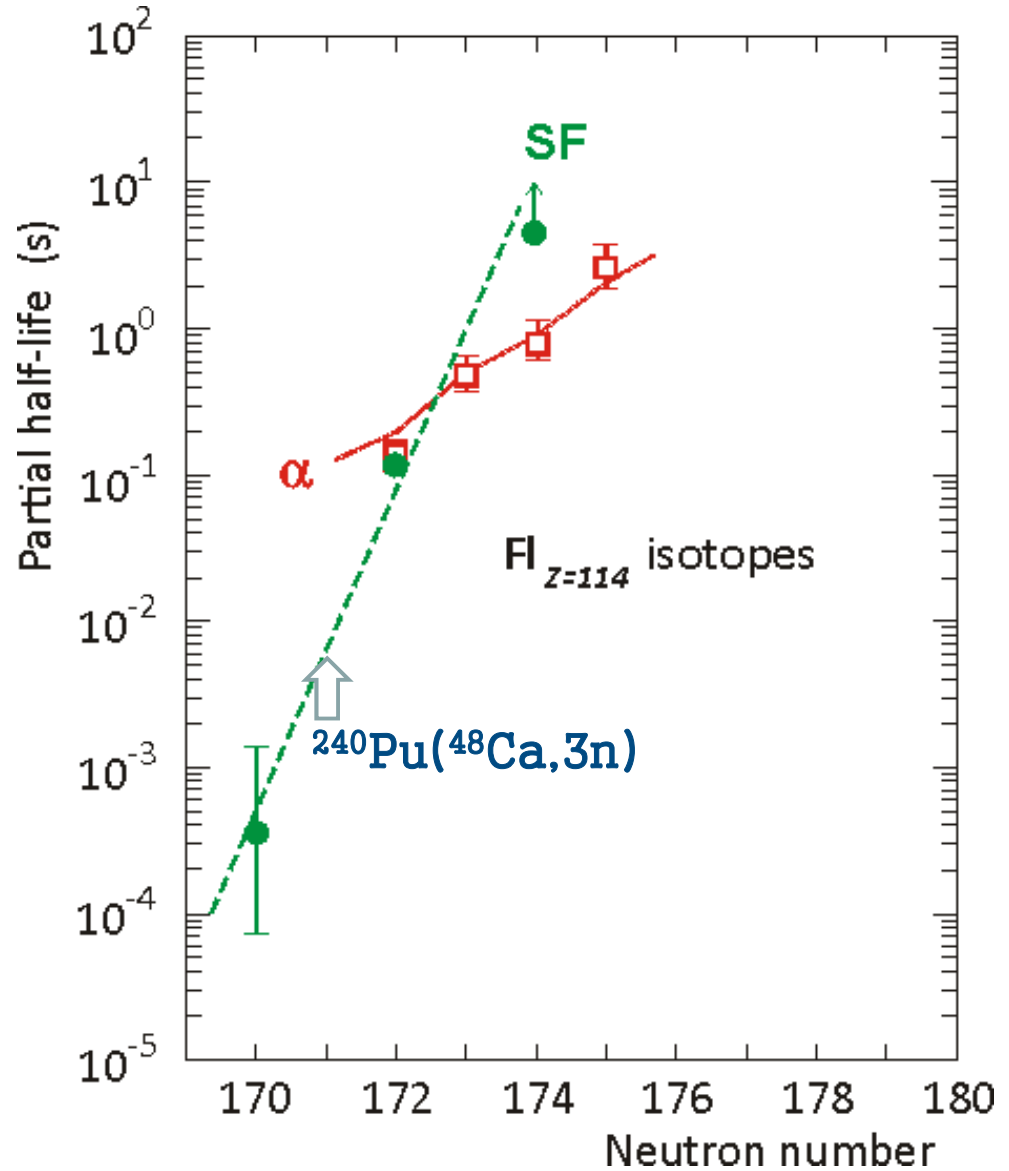
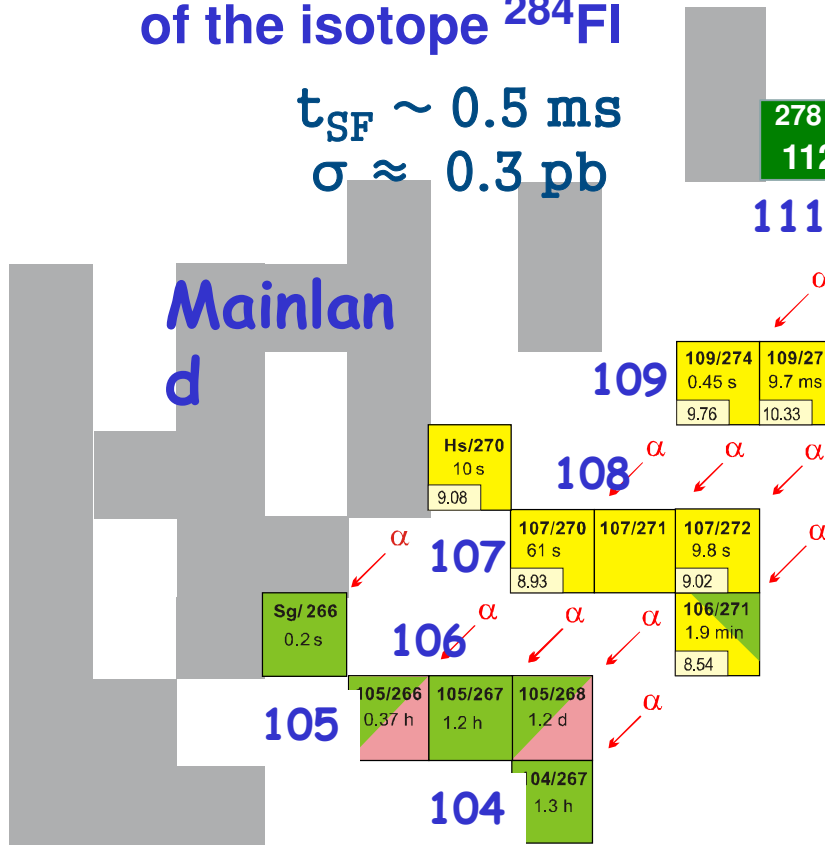
A/Z	Setup	Laboratory	Publications
$^{283}_{112}$	SHIP	GSI Darmstadt	Eur. Phys. J. A32, 251 (2007)
$^{283}_{112}$	COLD	PSI-FLNR (JINR)	NATURE 447, 72 (2007)
$^{286, 287}_{114}$	BGS	LBNL (Berkeley)	P.R. Lett. 103, 132502 (2009)
$^{288, 289}_{114}$	TASCA	GSI – Mainz	P.R. Lett. 104, 252701 (2010)
$^{292, 293}_{116}$	SHIP	GSI Darmstadt	Eur. Phys. J. A48, 62 (2012)
$^{287, 288}_{115}$	TASCA	GSI – Mainz	P.R. Lett. 111, 112502 (2013)
$^{293, 294}_{117}$	TASCA	GSI – Mainz	P.R. Lett. 112, 172501 (2014)
$^{292, 293}_{116}$	GARIS	RIKEN Tokyo	Accelerator Progress Rep. (2013)



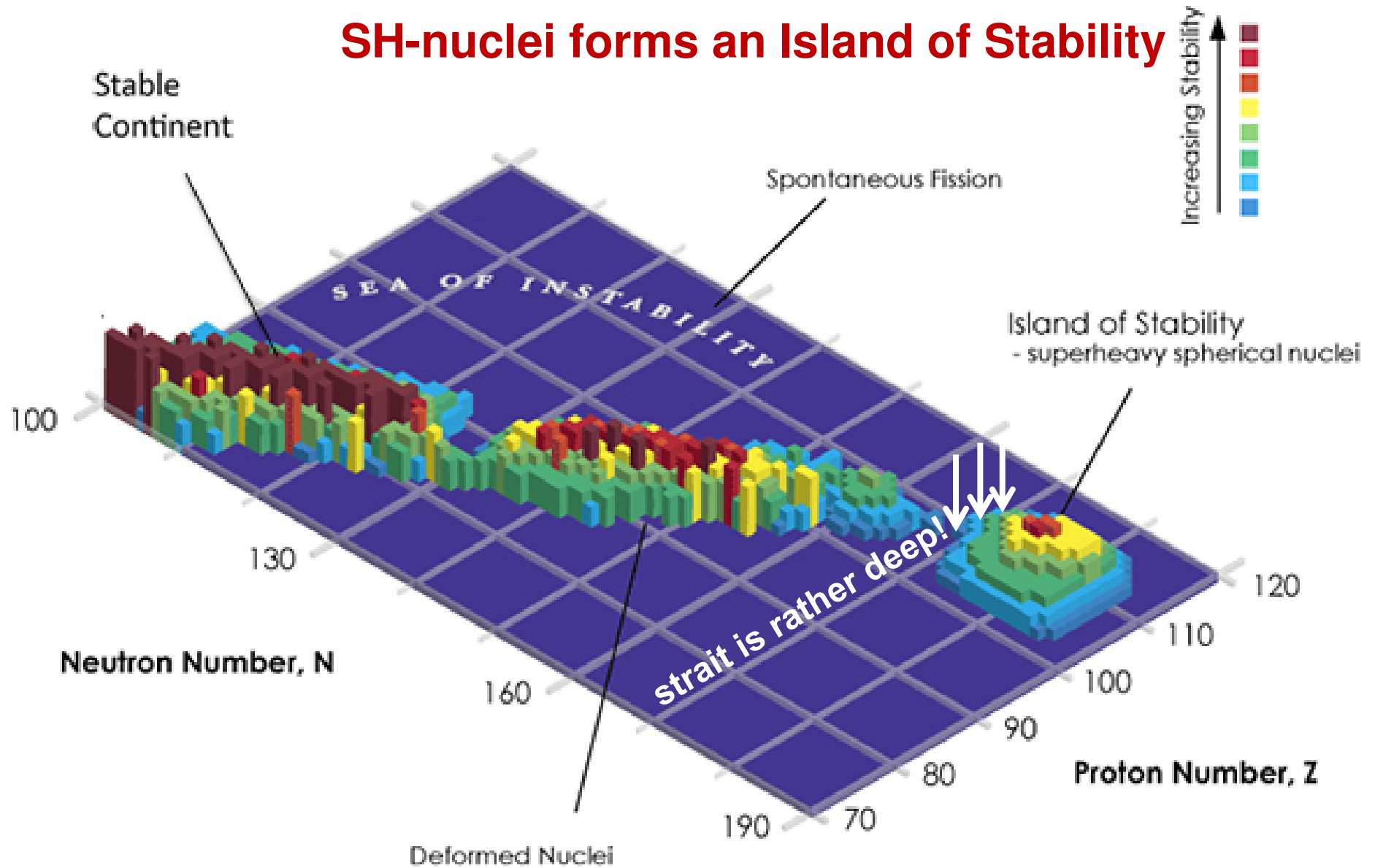


There were only one event,
as a candidate for the decay
of the isotope ^{284}Fl

$t_{\text{SF}} \sim 0.5 \text{ ms}$
 $\sigma \approx 0.3 \text{ pb}$



SH-nuclei forms an Island of Stability



With Z >40% larger than that of Bi, the heaviest stable element, we see an impressive extension in nuclear survival.

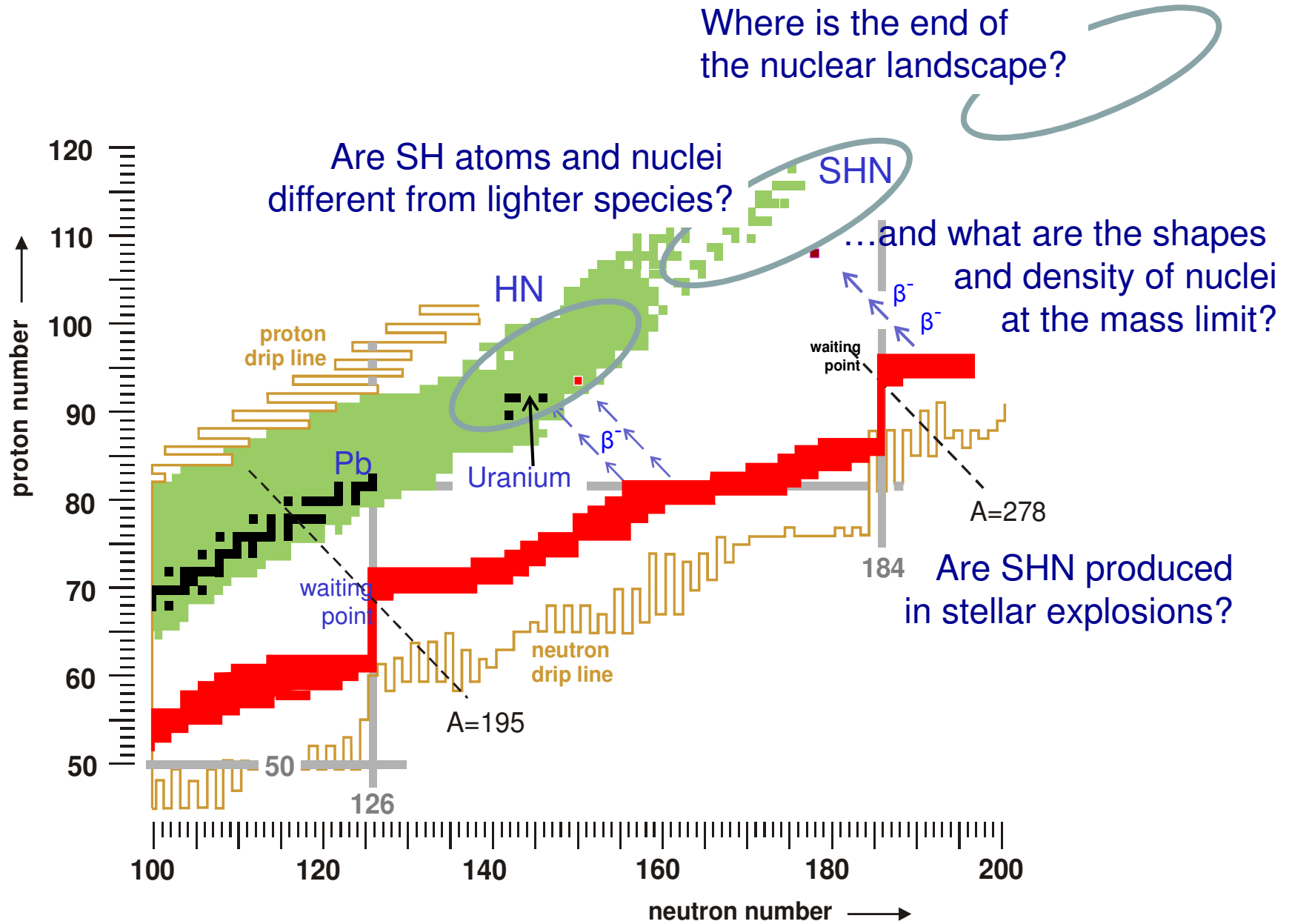
Although SHN are at the limits of Coulomb stability,

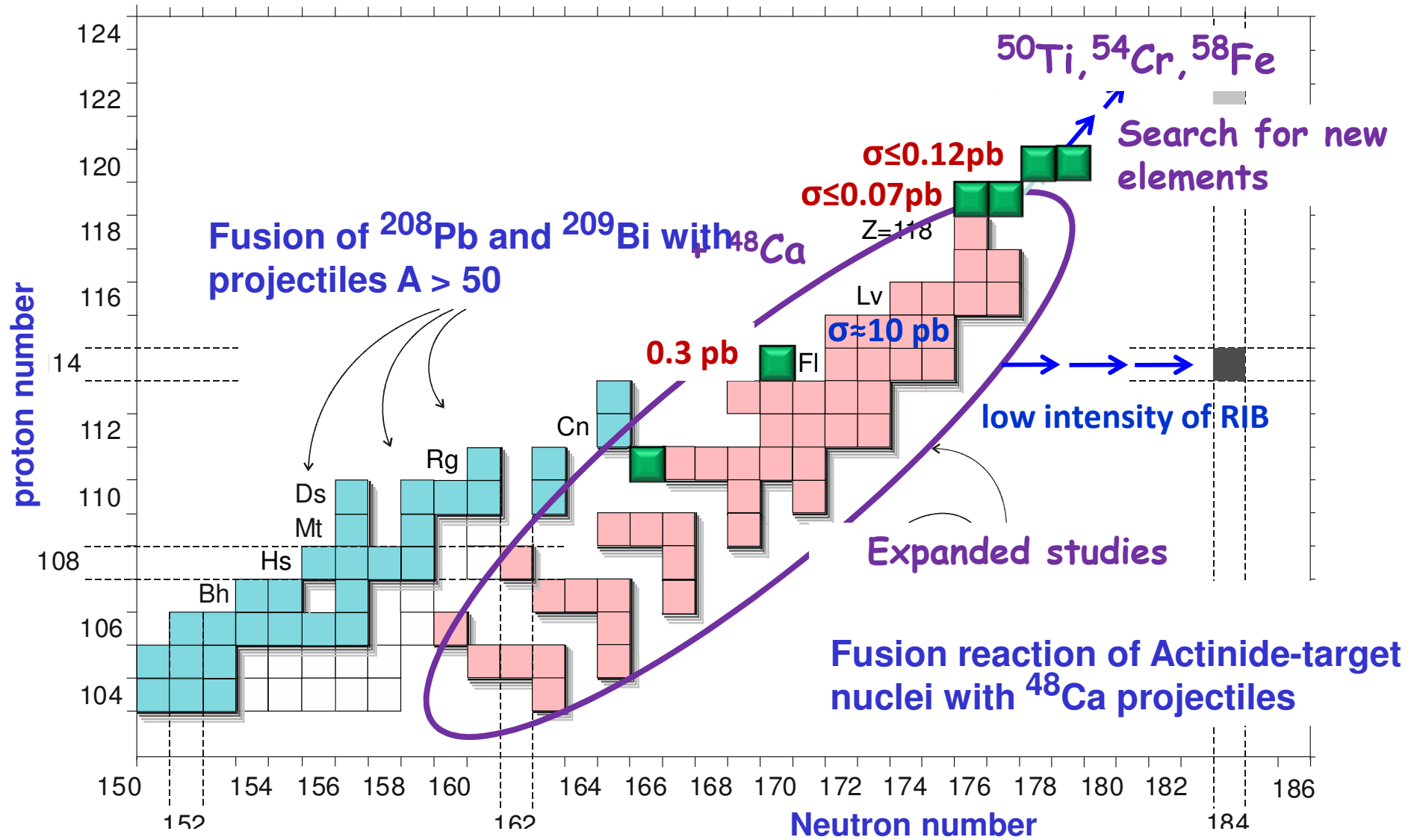
- shell stabilization lowers ground-state energy,
- creates a fission barrier,
- and thereby enables SHN to exist.

The fundamentals of the modern theory for mass limits of nuclear matter were given experimental verification.

**The discovery of SHE raised a
questions:**

**The discovery of SHE raised a
questions:**





Obviously...

**the field of the research is limited
by the production of super heavy nuclei**

Everything we know about SH-nuclei produced
in ^{48}Ca -induced reactions:

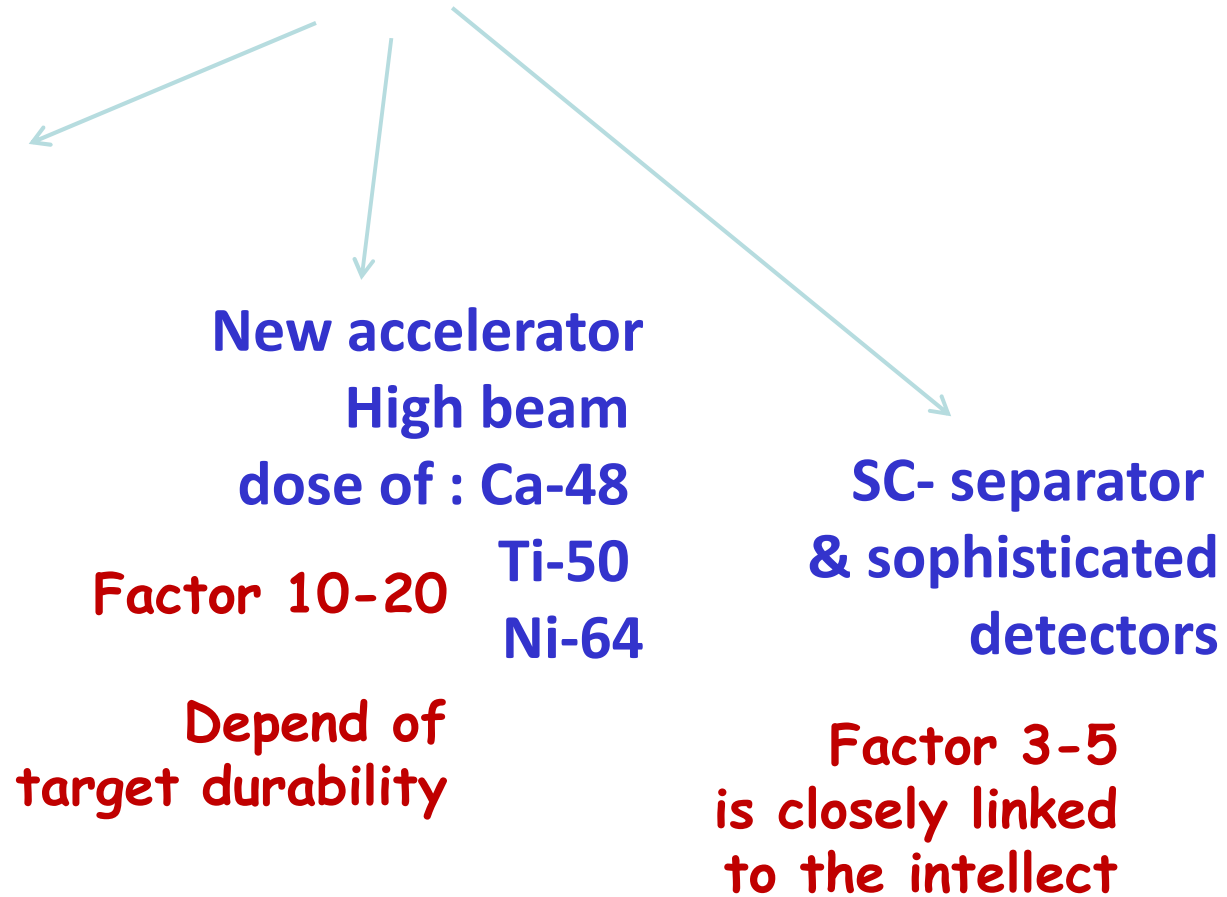
**...allow us to think about a SHE-Factory
with production rate about 100 times higher
than what we currently have**

Factory of SHE

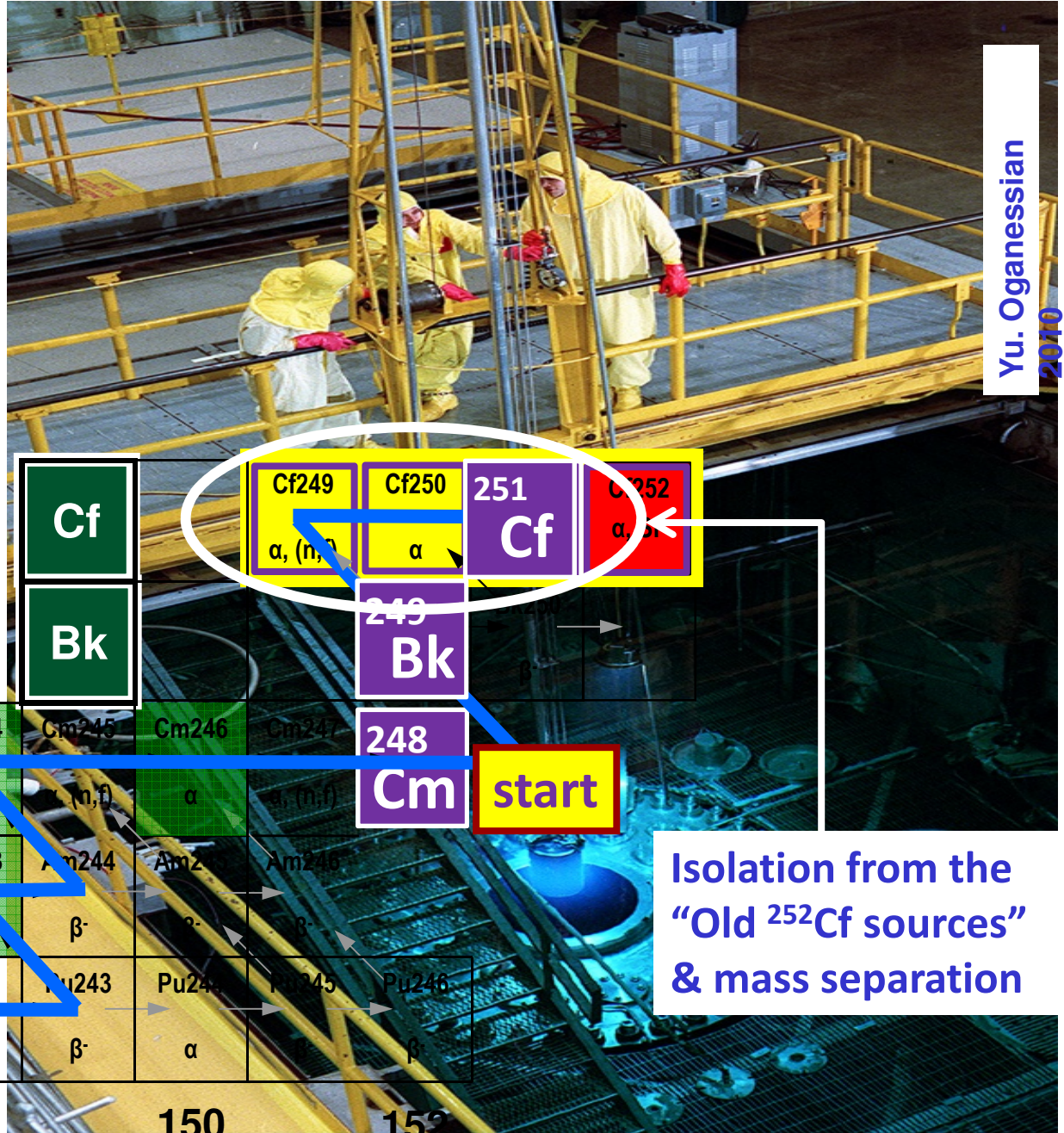
SHE-Factory

Isotope production:
Cm-248
Bk-249
Cf-251

To be increased
10 times

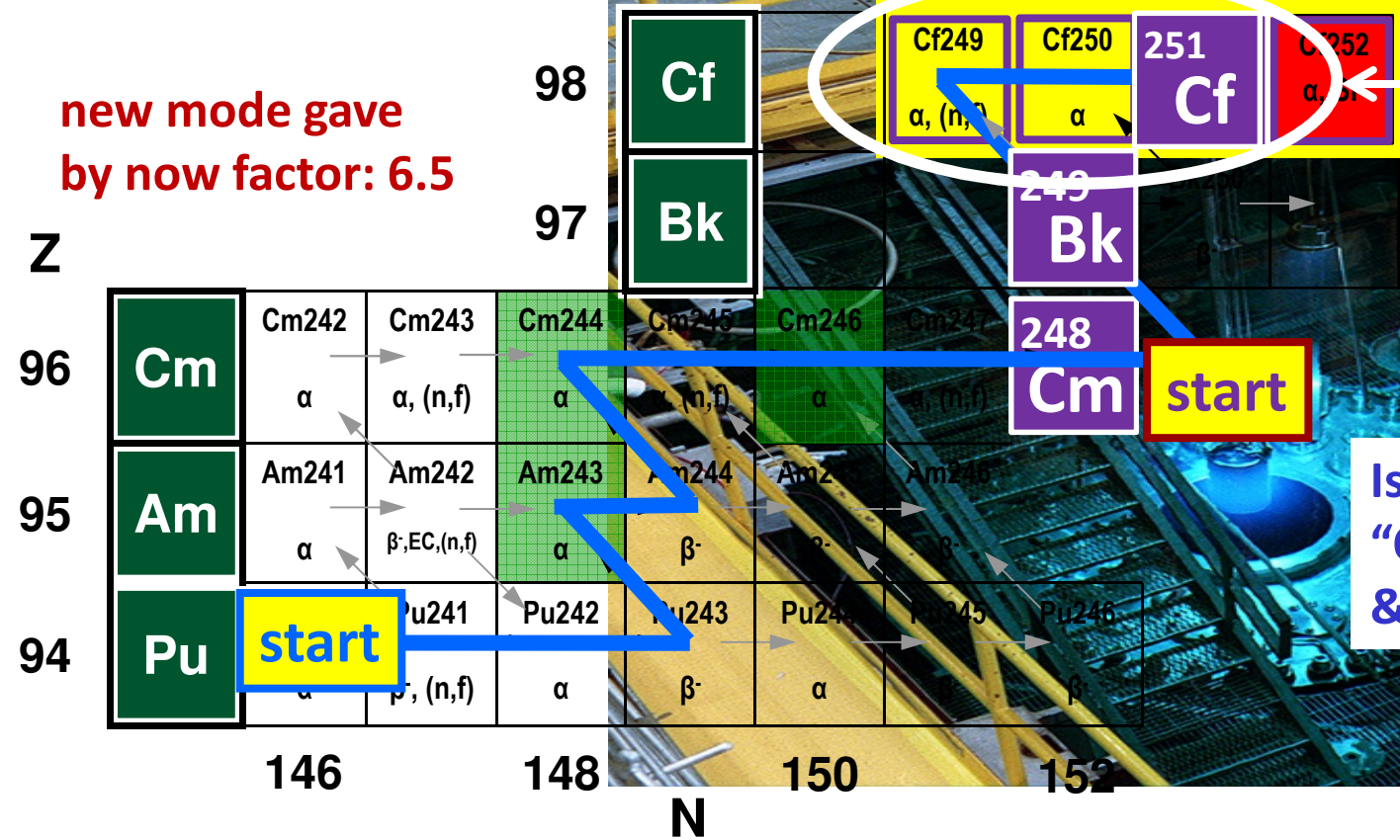


High Flux Isotope Reactor at Oak-Ridge



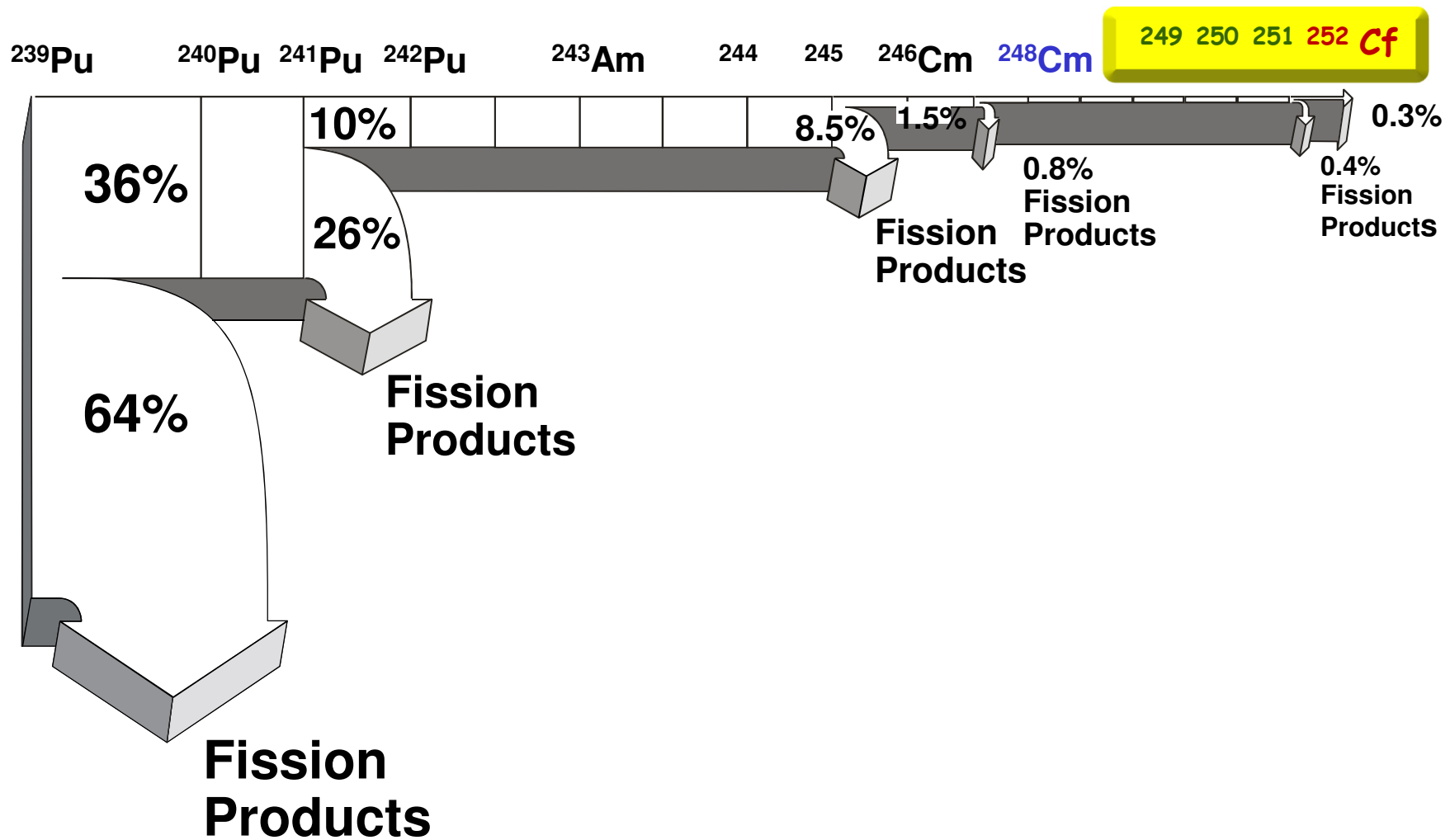
Yu. Oganessian 2010

new mode gave
by now factor: 6.5



Isolation from the
"Old ²⁵²Cf sources"
& mass separation

Fission Loss at Heavy Actinides





In solution

Cf chemical fraction at hot cell

April 2014

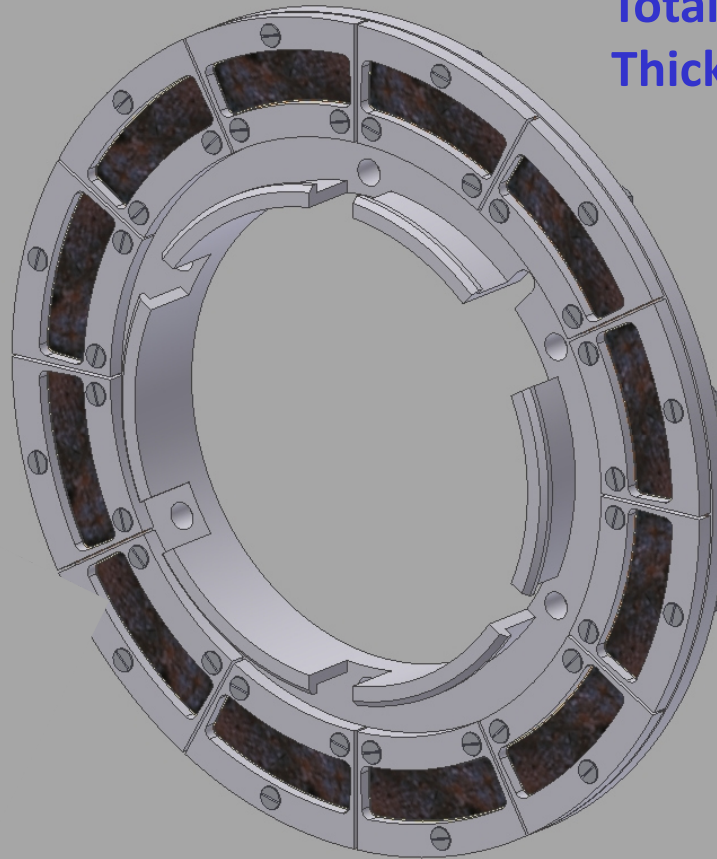
ORNL, Oak Ridge, Tennessee, USA

12 segments of rotational target made from mixed Cf-isotopes

Total weight 11.6 mg
Total surface 36 cm²
Thickness 0.32 mg/cm²

Segment

1/12



Cf-252	3 · 10 ⁻³ mg
Cf-251	4.19 mg
Cf-250	1.56 mg
Cf-249	5.84 mg

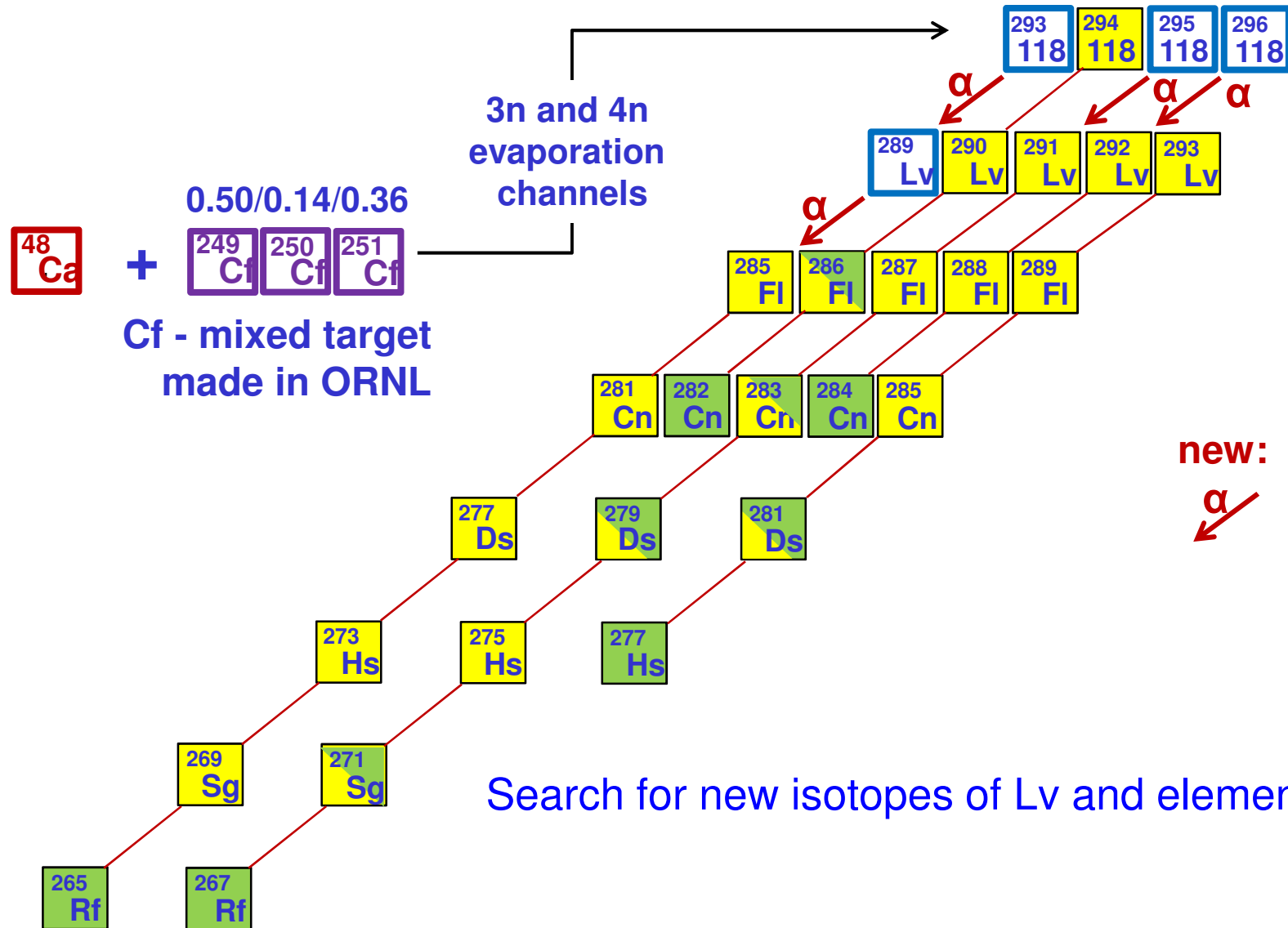
Oak Ridge National Laboratory, August, 2014

September 2014



Oak Ridge National Laboratory, USA

HEAVIEST NUCLEI



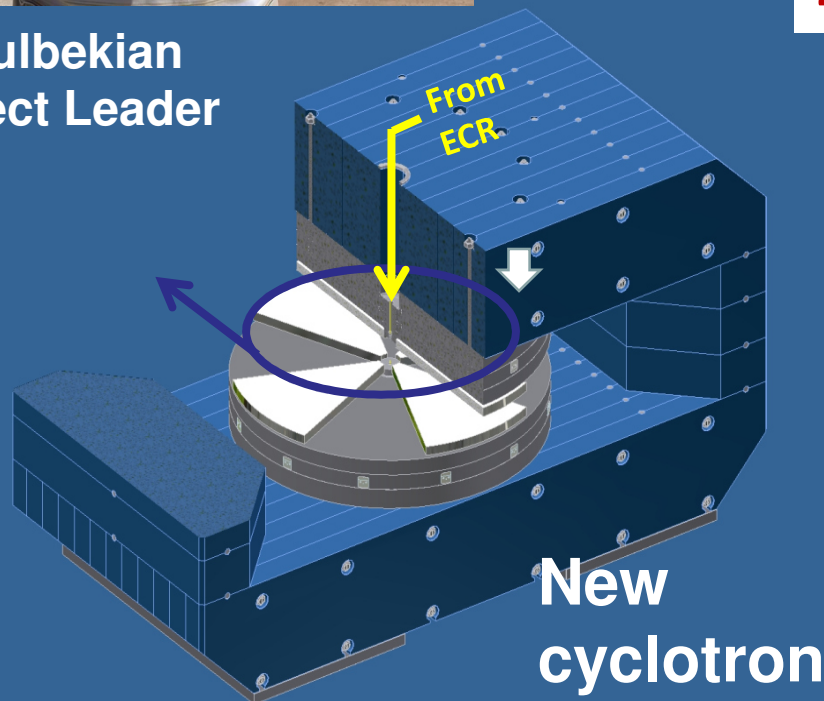


G. Gulbekian
Project Leader

New accelerator and new Lab. in Dubna

today: $\sim 5 \cdot 10^{19}/y$ with Factory: $1.0 \cdot 10^{21}/y$

	beam dose		factor: ~ 20
	Beam intensity	&	Beam time
	10-20 pμA		↓
			Factory
			↓
			~ 7000 h/year



**New
cyclotron**

July 2014, Dubna

new
accelerator

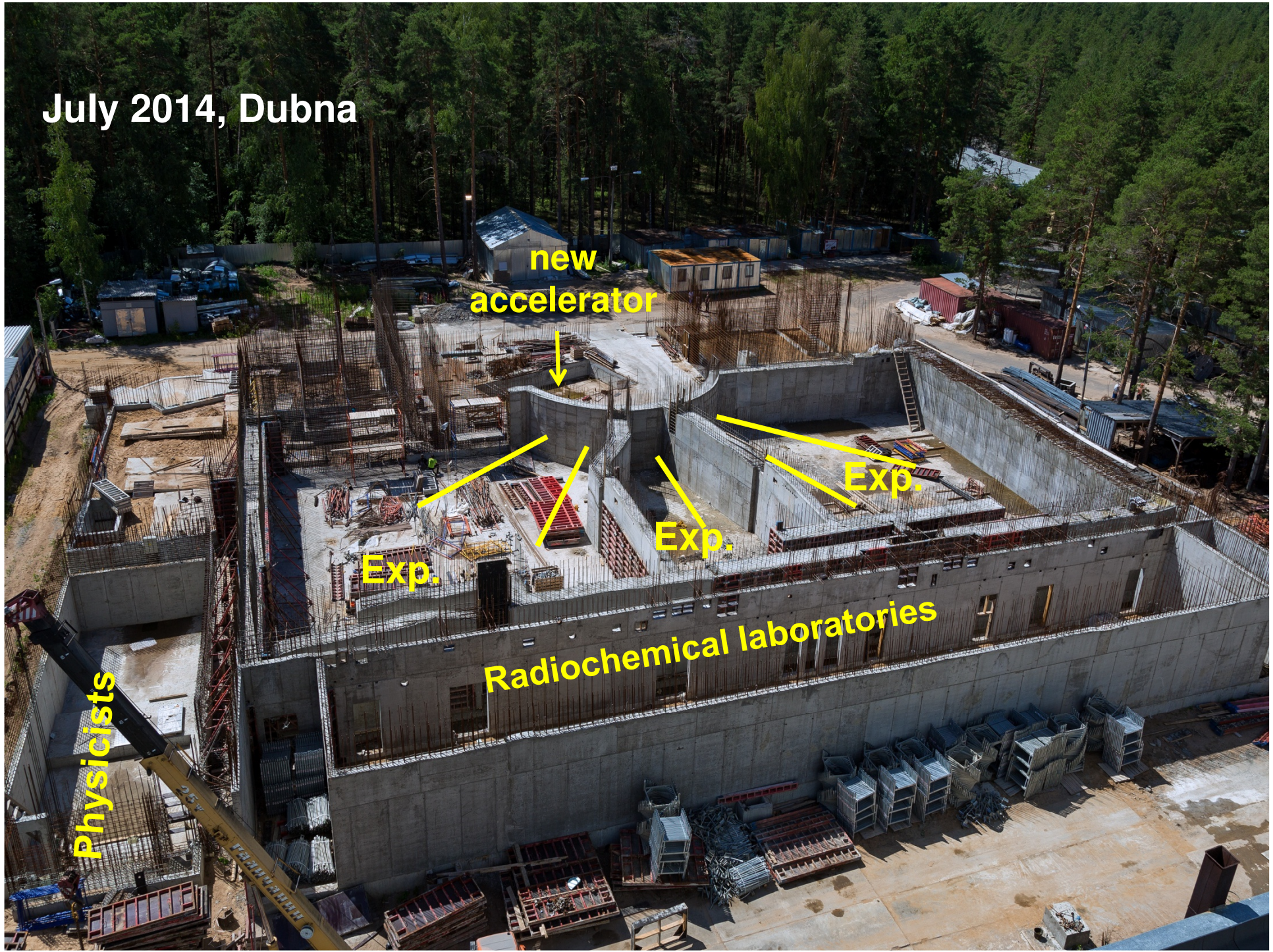
Exp.

Exp.

Exp.

Radiochemical laboratories


Physicists





August 2014, Dubna

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A large industrial facility, likely a machine shop or manufacturing plant, with a massive cylindrical component being processed. The component is mounted on a large, dark, rectangular base. The facility has a high ceiling with a corrugated metal roof and various structural elements. The lighting is bright, highlighting the metallic surfaces.

**Novokramatorks
Ukraine**

June 2014

Yu/ Oganessian. XXII International Baldin Seminar, September 15-20, 2014, JINR, Dubna

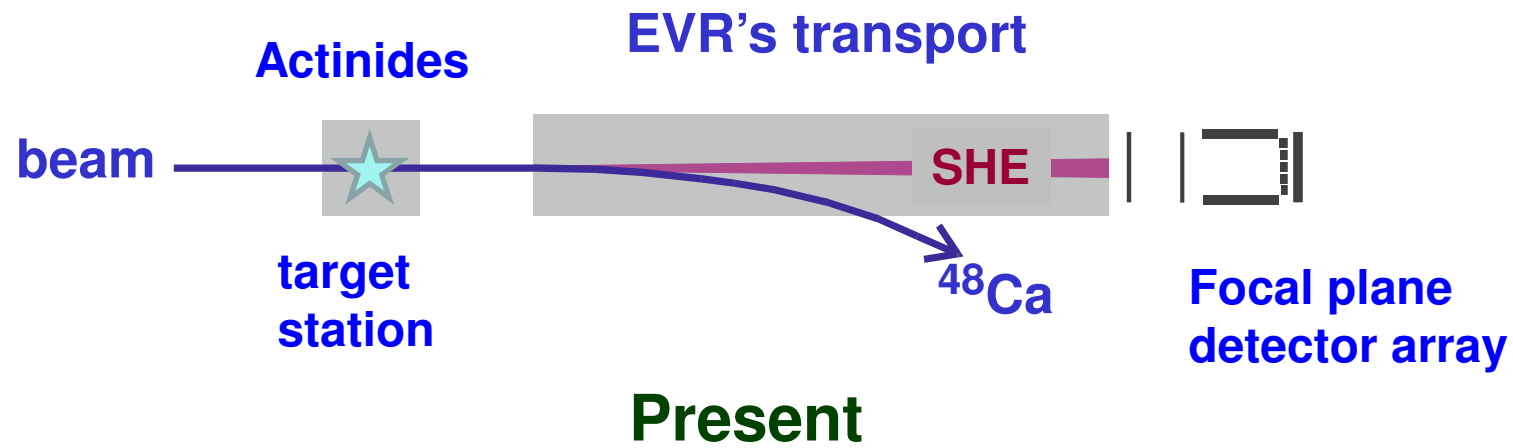


**Novokramatorsks
Ukraine**

August 2014

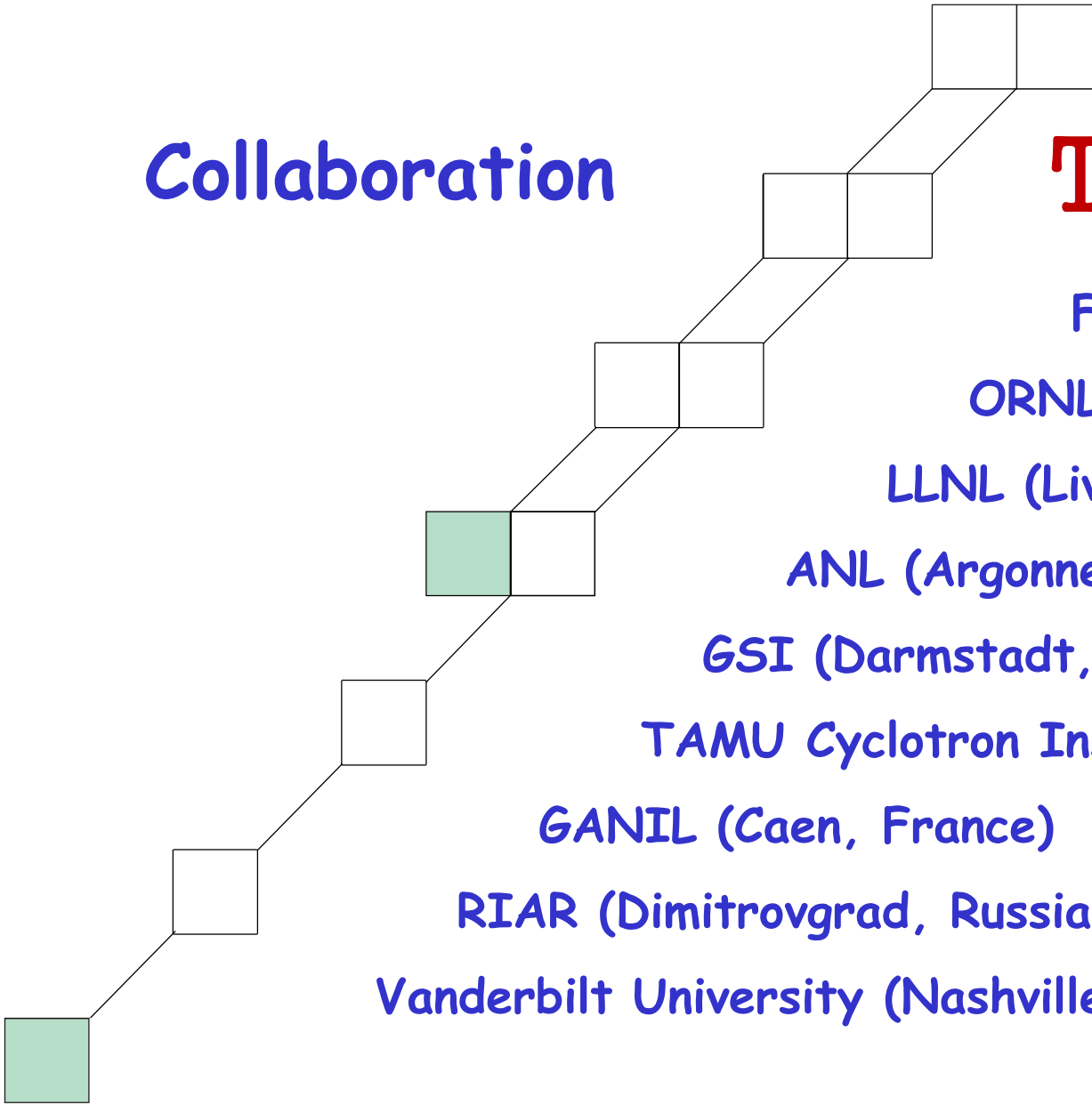
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Scheme of the production and delivery SH-atoms to the detectors



Collaboration

Thank you



FLNR, JINR (Dubna)

ORNL (Oak-Ridge, USA)

LLNL (Livermore, USA)

ANL (Argonne, USA)

GSI (Darmstadt, Germany)

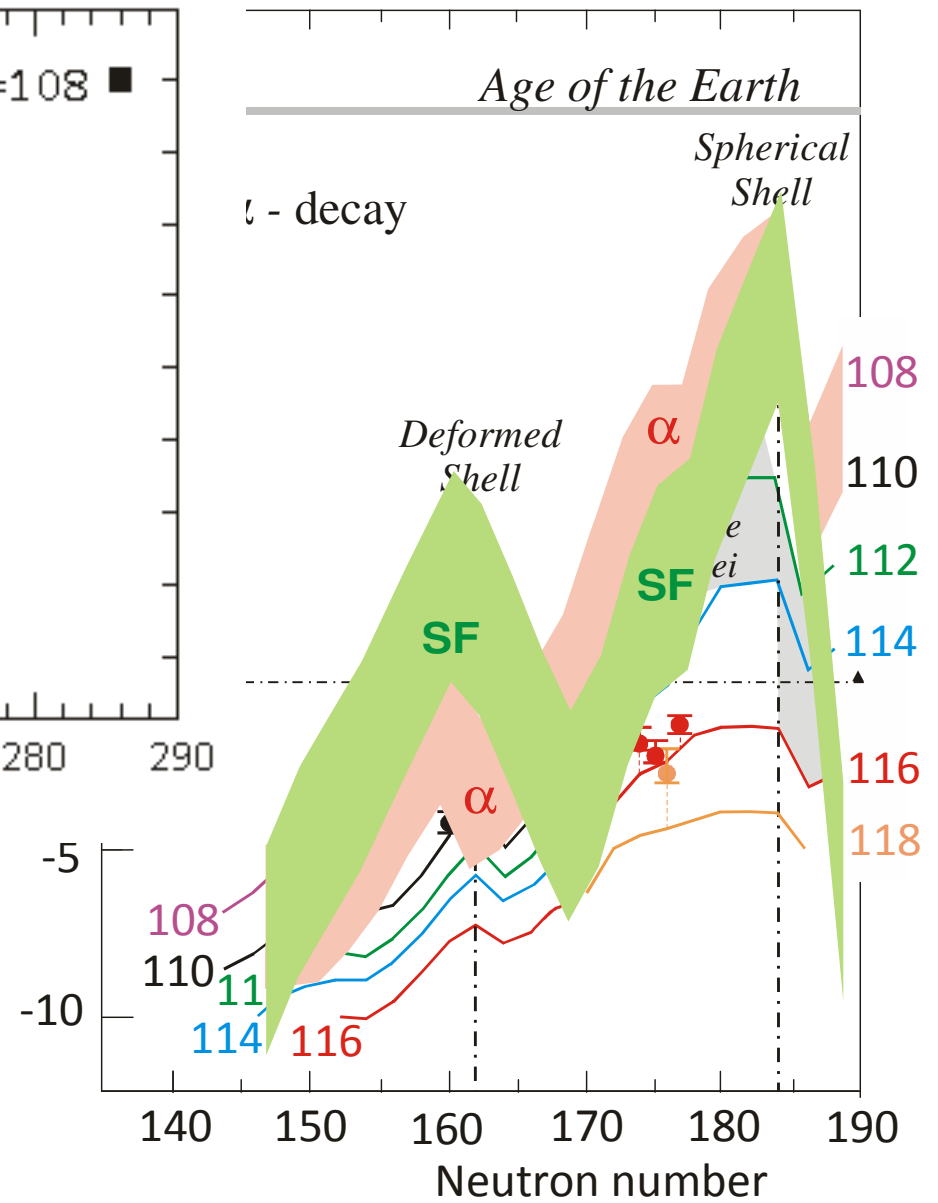
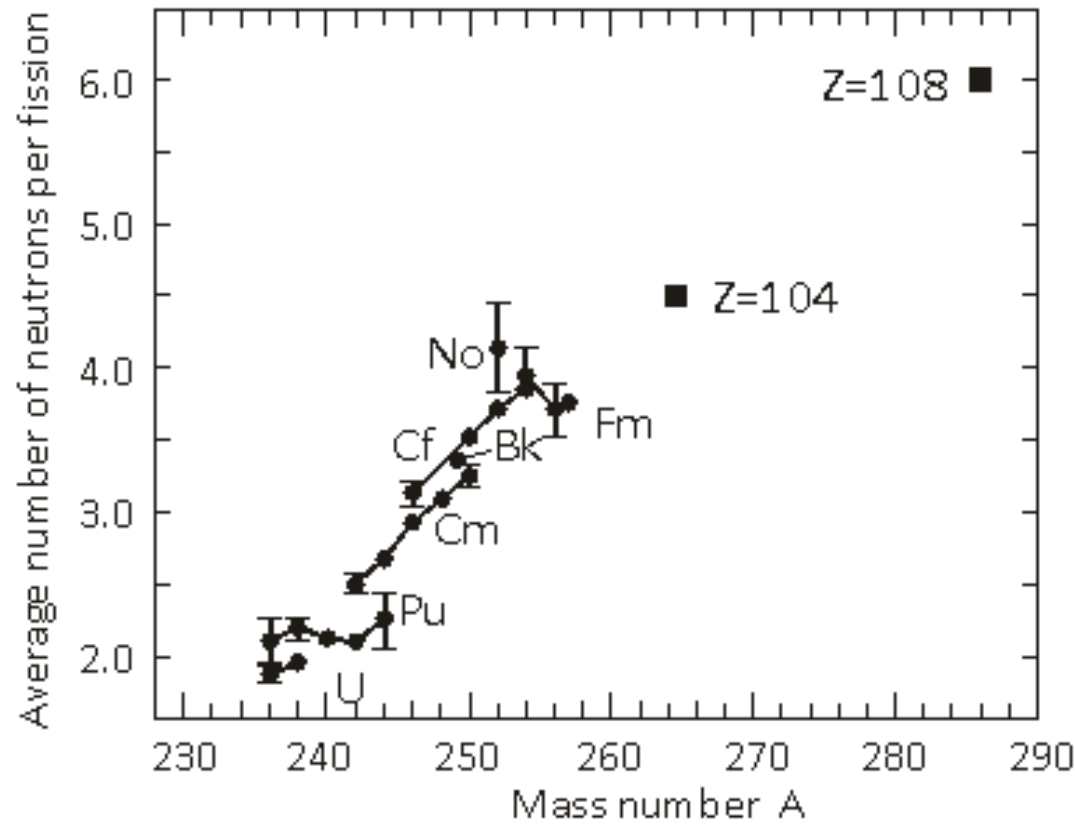
TAMU Cyclotron Institute (Texas, USA)

GANIL (Caen, France)

RIAR (Dimitrovgrad, Russia)

Vanderbilt University (Nashville, USA)

Theory and Experiment



Assuming for the SH-nuclide $T_{SF} = 10^9$ years

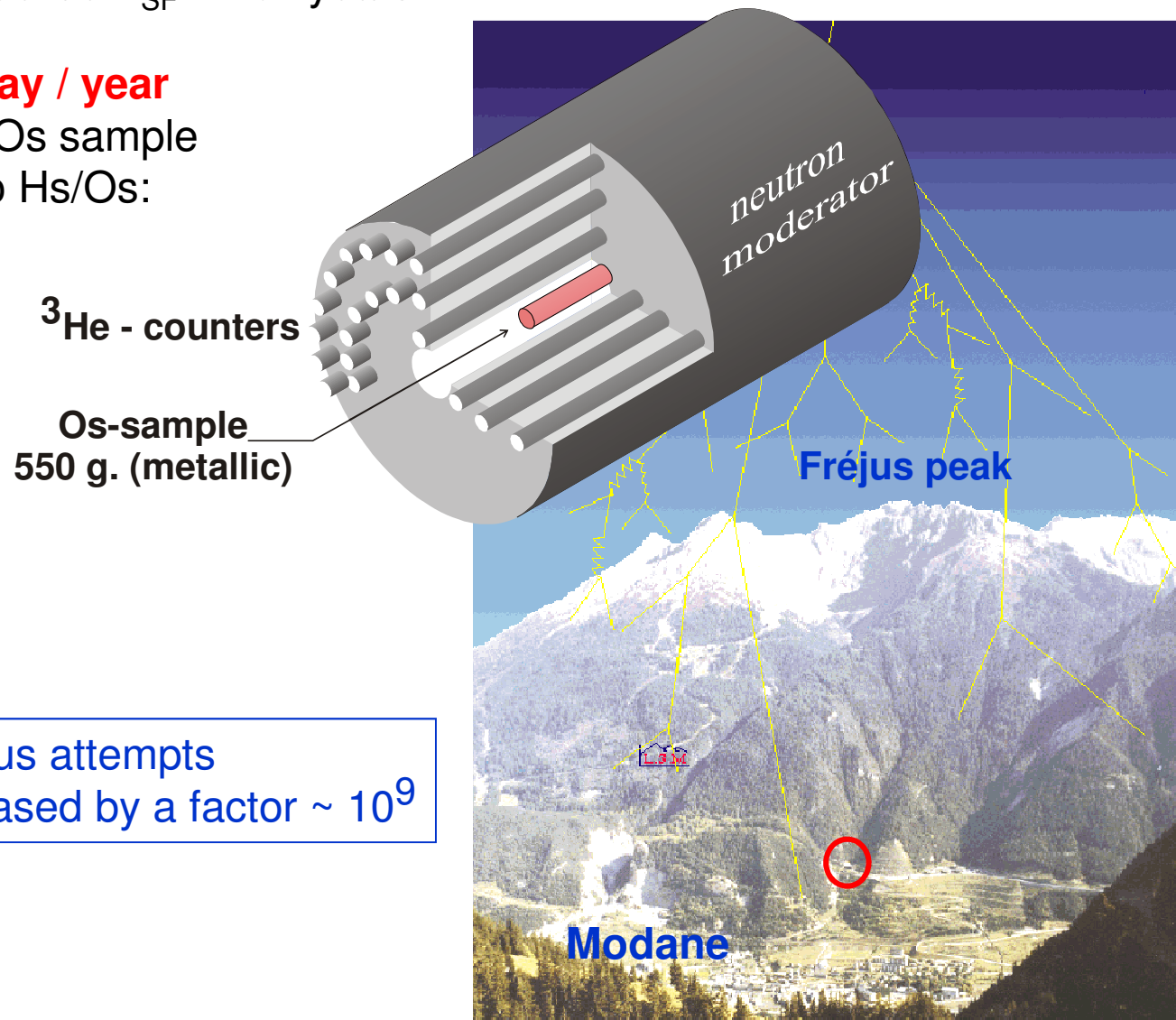
the counting rate **1 decay / year**
from a 1000-g metallic Os sample
corresponds to the ratio Hs/Os:

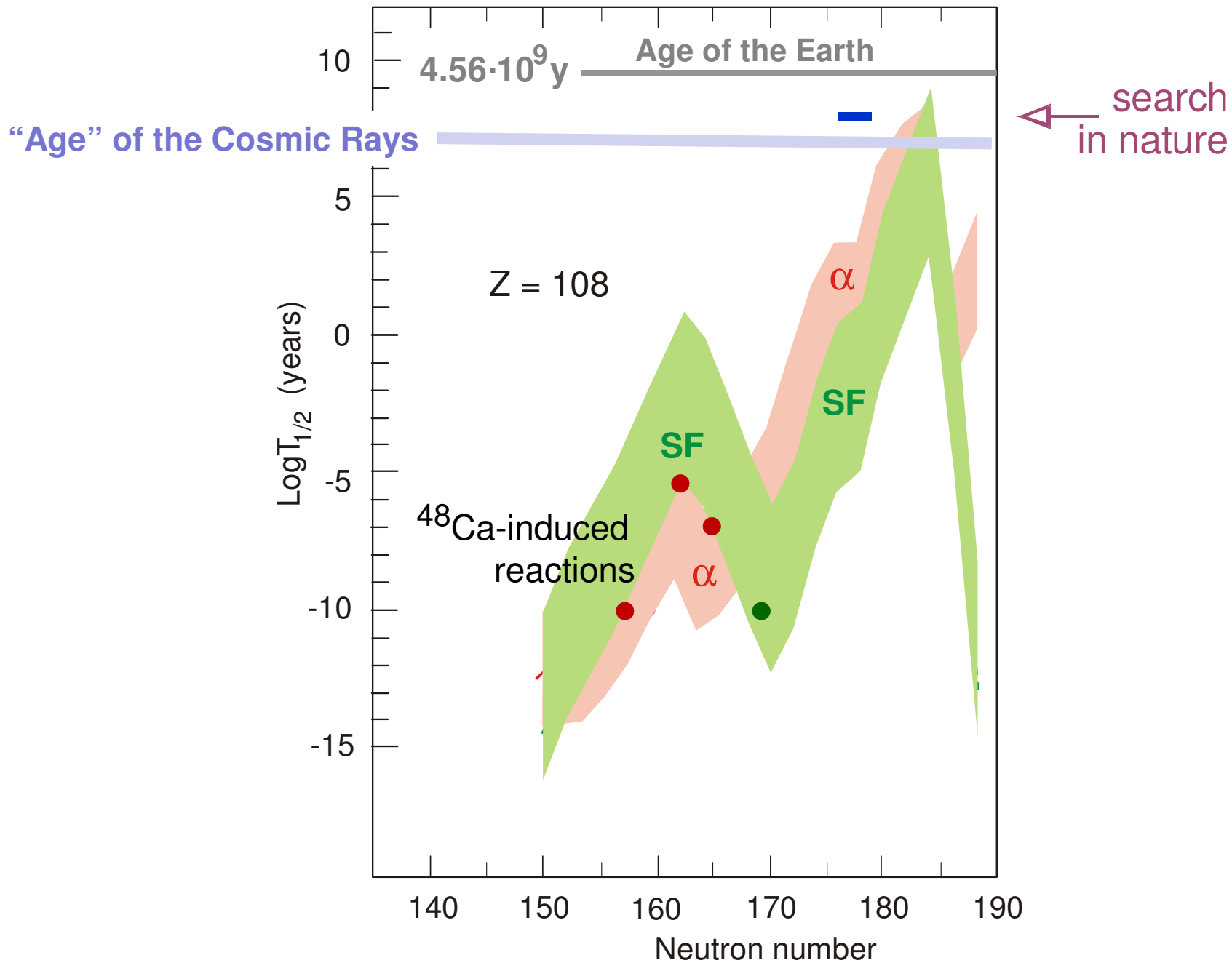
$$\sim 7 \cdot 10^{-16} \text{ g/g}$$

$$\text{or } \sim 10^{-23} \text{ g/g}$$

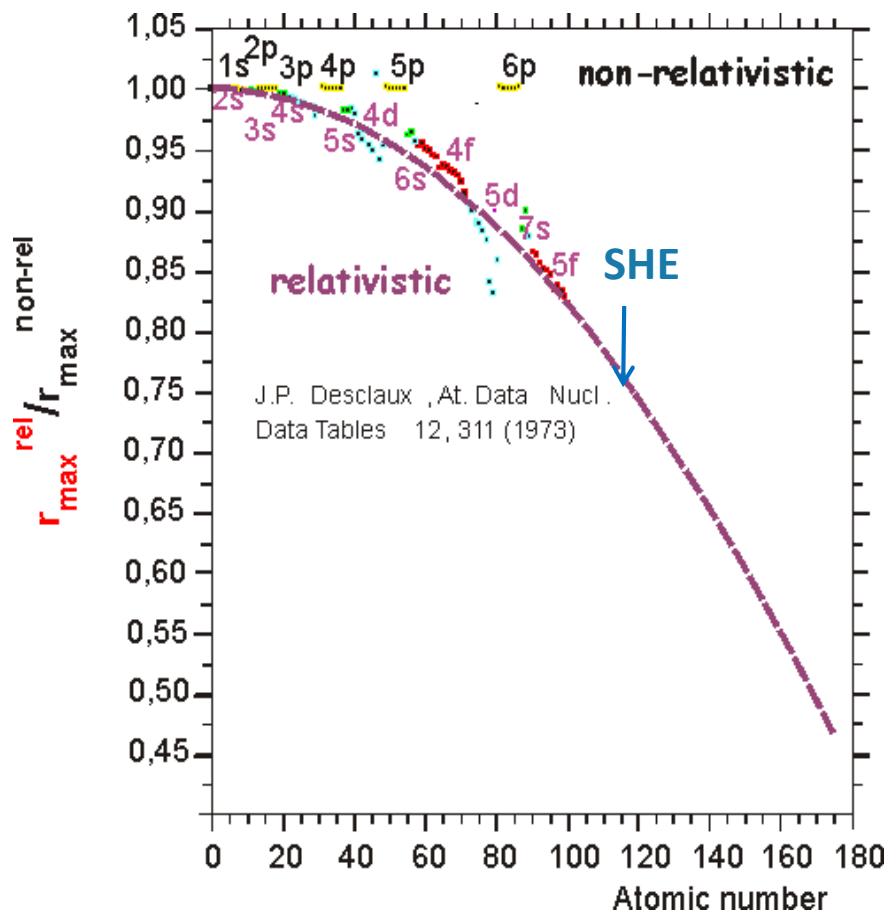
in the Earth's crust
or in the meteorit's
matter

compared with previous attempts
the sensitivity is increased by a factor $\sim 10^9$





Relativistic Contraction



Periodic Table

Z=1-172

Period	1	2											18 Orbitals						
1	1 H	2 He											13	14	15	16	17	18	1s
2	3 Li	4 Be											5	6	7	8	9	10	2s2p
3	11 Na	12 Mg	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	3s3p
4	19 K	20 Ca	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	4s3d4p
5	37 Rb	38 Sr	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	5s4d5p
6	55 Cs	56 Ba	57-71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	6s5d6p
7	87 Fr	88 Ra	89-103	104	105	106	107	108	109	110	111	112	113	114	115	116	117	118	7s6d7p
8	119	120	121-	156	157	158	159	160	161	162	163	164	165	166	167	168	169	170	8s7d8p
9	165	166																	9s9p

Inset tables:

6	LANTANIDES										4f					
7	ACTINIDES										5f					
8	141	142	143	144	145	146	147	148	149	150	151	152	153	154	155	6f

8	121	122	123	124	125	126	127	128	129	130	131	132	133	134	135	136	137	138	5g
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Periodic Table based on Dirac-Fock calculations (non-relativistic)

Periodic Table of Elements

1											13	14	15	16	17	18		
1	H 1 Hydrogen											B 5 Boron	C 6 Carbon	N 7 Nitrogen	O 8 Oxygen	F 9 Fluorine	He 2 Helium	
2	Li 3 Lithium	Be 4 Beryllium											Al 13 Aluminum	Si 14 Silicon	P 15 Phosphorus	S 16 Sulfur	Cl 17 Chlorine	Ne 10 Neon
3	Na 11 Sodium	Mg 12 Magnesium											Ga 31 Gallium	Ge 32 Germanium	As 33 Arsenic	Se 34 Selenium	Br 35 Bromine	Ar 18 Argon
4	K 19 Potassium	Ca 20 Calcium	Sc 21 Scandium	Ti 22 Titanium	V 23 Vanadium	Cr 24 Chromium	Mn 25 Manganese	Fe 26 Iron	Co 27 Cobalt	Ni 28 Nickel	Cu 29 Copper	Zn 30 Zinc	In 49 Indium	Sn 50 Tin	Sb 51 Antimony	Tc 52 Technetium	I 53 Iodine	Kr 36 Krypton
5	Rb 37 Rubidium	Sr 38 Strontium	Y 39 Yttrium	Zr 40 Zirconium	Nb 41 Niobium	Mo 42 Molybdenum	Tc 43 Technetium	Ru 44 Ruthenium	Rh 45 Rhodium	Pd 46 Palladium	Ag 47 Silver	Cd 48 Cadmium	Hg 80 Mercury	Pb 82 Lead	Bi 83 Bismuth	Po 84 Polonium	At 85 Astatine	Xe 54 Xenon
6	Cs 55 Cesium	Ba 56 Barium	La 57 Lanthanum	Hf 72 Hafnium	Ta 73 Tantalum	W 74 Tungsten	Re 75 Rhenium	Os 76 Osmium	Ir 77 Iridium	Pt 78 Platinum	Au 79 Gold	Hg 80 Mercury	Tl 81 Thallium	Pb 82 Lead	Bi 83 Bismuth	Po 84 Polonium	At 85 Astatine	Rn 86 Radon
7	Fr 87 Francium	Ra 88 Radium	Ac 89 Actinium	Rf 104 Rutherfordium	Db 105 Dubnium	Sg 106 Seaborgium	Bh 107 Bohrium	Hs 108 Hassium	Mt 109 Meitnerium	Ds 110 Darmstadtium	Rg 111 Roentgenium	112	113	114	115	116	117	118

➡ more and more inert?

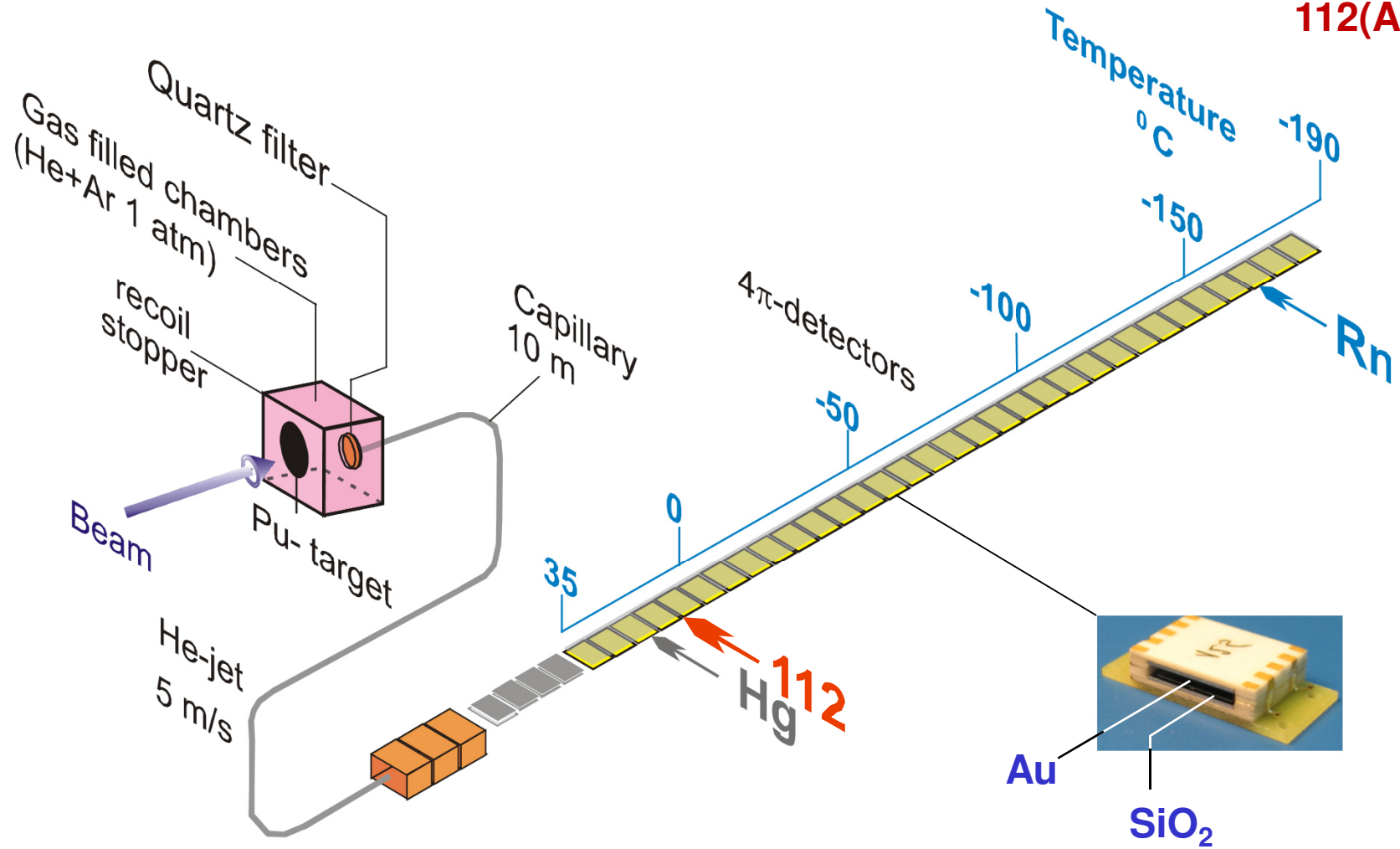
Lanthanides	La 57 Lanthanum	Ce 58 Cerium	Pr 59 Praseodymium	Nd 60 Neodymium	Pm 61 Promethium	Sm 62 Samarium	Eu 63 Europium	Gd 64 Gadolinium	Tb 65 Terbium	Dy 66 Dysprosium	Ho 67 Holmium	Er 68 Erbium	Tm 69 Thulium	Yb 70 Ytterbium	Lu 71 Lutetium
Actinides	Ac 89 Actinium	Th 90 Thorium	Pa 91 Protactinium	U 92 Uranium	Np 93 Neptunium	Pu 94 Plutonium	Am 95 Americium	Cm 96 Curium	Bk 97 Berkeleium	Cf 98 Californium	Es 99 Einsteinium	Fm 100 Fermium	Md 101 Mendelevium	No 102 Nobelium	Lr 103 Lawrencium

Yu. Oganessian. "At the End of the Periodic Table" 2013 SCNAT Annual Congress, Nov.22, Winterthur, Switzerland

Reaction:



Compounds:
Hg(Au)
and
112(Au)



Element 112 is a noble metal – like Hg!

